

INTRODUCTION TO HUMAN PHYSIOLOGY – CHEMISTRY BASICS REMINDERS

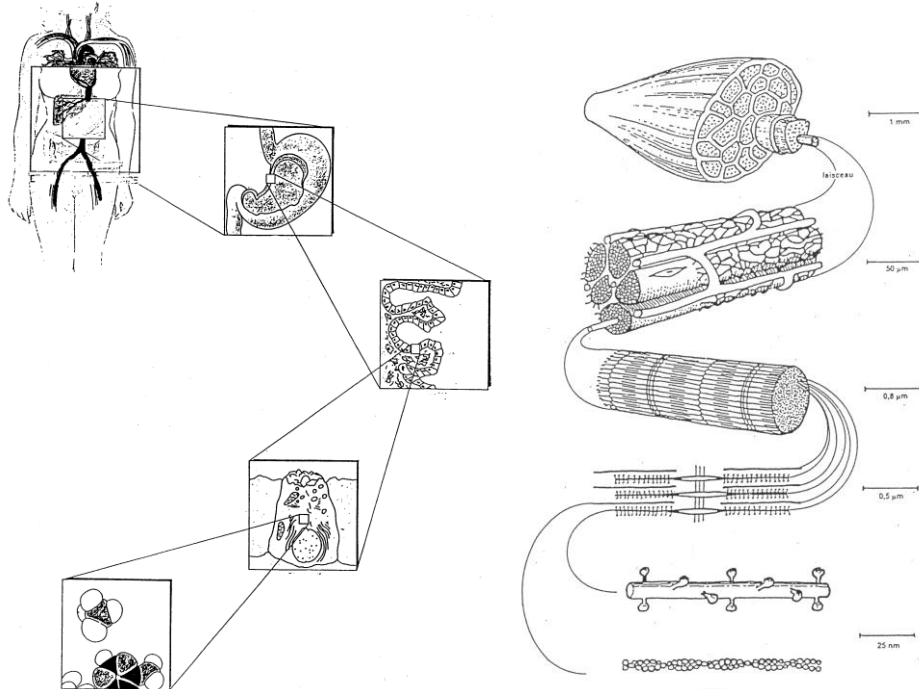
EXERCISE 1

- 1- Give a definition of human physiology
- 2- Reorder the main systems of the human organism from the most involved to the less involved when performing a physical exercise :

Cardiovascular system / Nervous system/ Skeletal system / Genital system / Muscular system / Endocrine system / Lymphatic and immune system / Integumentary system / Digestive system / Respiratory system / Urinary system

EXERCISE 2

Give the levels of organization of the 2 figures below:



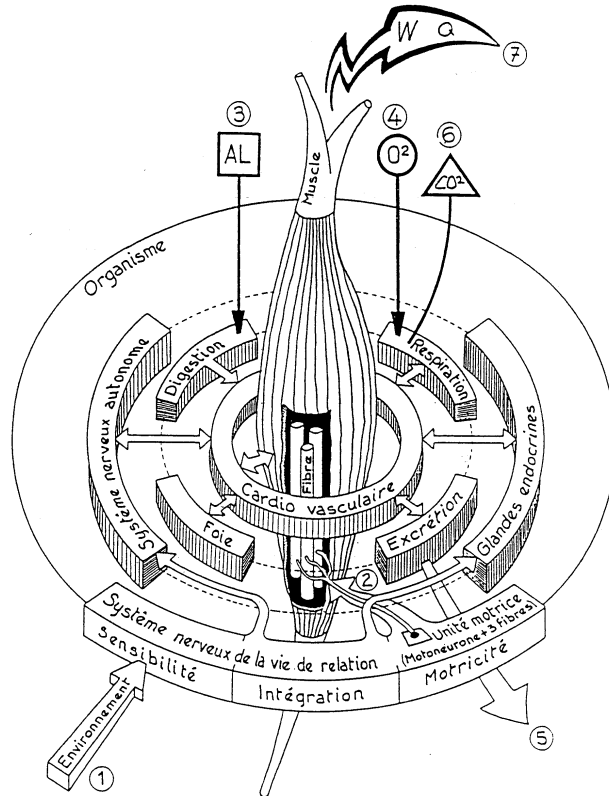
EXERCISE 3

When exercising, some common physiological adaptations/modulations could commonly occur and influence the performance. List some of these events and name the tissues/organs/system that are involved (there is no “right” answer since many adaptations can occur during exercise, just try to be logical by thinking to the most common ones).

EXERCISE 4

Give a title to the figure below. Thereafter, please discuss the three following points:

- meaning of this circular representation
- role of each main function (roughly)
- meaning of arrows from 1 to 7

**EXERCISE 5**

- 1- Give the list of the main atoms that are present in the human body
- 2- By using the periodic table (see below), draw the atomic orbital of the Carbon and Sodium atoms
- 3- Describe the two types of chemical bonds that are the most common between atoms (*ionic and covalent bonds*). Give an example for each of them (orbital drawing, developed formula and simple formula for the chosen molecule). **Take Na & Cl atoms for the example of ionic bond, and H & O atoms for the example of covalent bond.**

<p>1 H Numéro atomique Hydrogène Symbôle 1,00794 Nom de l'élément Masse atomique</p>																<p>18 He Hélium 4,00</p>															
1 H Hydrogène 1,00794															18 He Hélium 4,00																
3 Li Lithium 6,94	4 Be Béryllium 9,01													5 B Bore 10,81	6 C Carbone 12,01	7 N Azote 14,01	8 O Oxygène 16,00	9 F Fluor 19,00	10 Ne Neon 20,18												
11 Na Sodium 22,99	12 Mg Magnésium 24,31													13 Al Aluminium 26,98	14 Si Silicium 28,09	15 P Phosphore 30,97	16 S Soufre 32,06	17 Cl Chlore 34,45	18 Ar Argon 39,95												
19 K Potassium 39,10	20 Ca Calcium 40,08	21 Sc Scandium 44,96	22 Ti Titane 47,88	23 V Vanadium 50,94	24 Cr Chrome 52,00	25 Mn Manganèse 54,94	26 Fe Fer 55,85	27 Co Cobalt 58,93	28 Ni Nickel 58,71	29 Cu Cuivre 63,54	30 Zn Zinc 65,37	31 Ga Gallium 69,72	32 Ge Germanium 72,59	33 As Arsenic 74,92	34 Se Sélénium 78,96	35 Br Brome 79,91	36 Kr Krypton 83,80														
37 Rb Rubidium 85,47	38 Sr Strontium 87,62	39 Y Yttrium 88,91	40 Zr Zirconium 91,22	41 Nb Niobium 92,91	42 Mo Molybdène 95,94	43 Tc Technétium 98,91	44 Ru Ruthénium 101,07	45 Rh Rhodium 102,91	46 Pd Paladium 106,4	47 Ag Argent 107,87	48 Cd Cadmium 112,40	49 In Indium 114,82	50 Sn Étain 118,69	51 Sb Antimoine 121,75	52 Te Tellure 127,60	53 I Iode 126,90	54 Xe Xénon 131,30														
55 Cs Césium 132,91	56 Ba Baryum 137,33	71 Lu Lutécium 174,97	72 Hf Hafnium 178,49	73 Ta Tantale 180,95	74 W Tungstène 183,85	75 Re Rhenium 186,2	76 Os Osmium 190,2	77 Ir Iridium 192,2	78 Pt Platine 195,09	79 Au Or 196,97	80 Hg Mercure 200,59	81 Tl Thallium 204,37	82 Pb Plomb 207,19	83 Bi Bismuth 208,98	84 Po Polonium 210	85 At Astaté 210	86 Rn Radon 222														
87 Fr Francium 223	88 Ra Radium 226,03	103 Lr Lawrencium 262,1	104 Rf Rutherfordium	105 Db Dubnium	106 Sg Seaborgium	107 Bh Bohrium	108 Hs Hassium	109 Mt Meitnerium	110 Uun	111 Uuu	112 Uub	113 Uut																			
<p>*Lanthanides</p> <table border="1"> <tr> <td>57 La* Lanthane 138,91</td> <td>58 Ce Cérium 140,12</td> <td>59 Pr Praseodyme 140,91</td> <td>60 Nd Néodyme 144,24</td> <td>61 Pm Prométhium 146,92</td> <td>62 Sm Samarium 150,35</td> <td>63 Eu Europium 151,96</td> <td>64 Gd Gadolinium 157,25</td> <td>65 Tb Terbium 158,92</td> <td>66 Dy Dysprosium 162,50</td> <td>67 Ho Holmium 164,93</td> <td>68 Er Erbium 167,26</td> <td>69 Tm Thulium 168,93</td> <td>70 Yb Ytterbium 173,04</td> </tr> </table>																		57 La* Lanthane 138,91	58 Ce Cérium 140,12	59 Pr Praseodyme 140,91	60 Nd Néodyme 144,24	61 Pm Prométhium 146,92	62 Sm Samarium 150,35	63 Eu Europium 151,96	64 Gd Gadolinium 157,25	65 Tb Terbium 158,92	66 Dy Dysprosium 162,50	67 Ho Holmium 164,93	68 Er Erbium 167,26	69 Tm Thulium 168,93	70 Yb Ytterbium 173,04
57 La* Lanthane 138,91	58 Ce Cérium 140,12	59 Pr Praseodyme 140,91	60 Nd Néodyme 144,24	61 Pm Prométhium 146,92	62 Sm Samarium 150,35	63 Eu Europium 151,96	64 Gd Gadolinium 157,25	65 Tb Terbium 158,92	66 Dy Dysprosium 162,50	67 Ho Holmium 164,93	68 Er Erbium 167,26	69 Tm Thulium 168,93	70 Yb Ytterbium 173,04																		
<p>**Actinides</p> <table border="1"> <tr> <td>89 Ac** Actinium 227,03</td> <td>90 Th Thorium 232,04</td> <td>91 Pa Protactinium 231,04</td> <td>92 U Uranium 238,03</td> <td>93 Np Neptunium 237,05</td> <td>94 Pu Plutonium 239,05</td> <td>95 Am Américium 241,06</td> <td>96 Cm Curium 247,07</td> <td>97 Bk Berkélium 249,08</td> <td>98 Cf Californium 251,08</td> <td>99 Es Einsteinium 254,09</td> <td>100 Fm Fermium 257,10</td> <td>101 Md Mendelevium 258,10</td> <td>102 No Nobelium 255</td> </tr> </table>																		89 Ac** Actinium 227,03	90 Th Thorium 232,04	91 Pa Protactinium 231,04	92 U Uranium 238,03	93 Np Neptunium 237,05	94 Pu Plutonium 239,05	95 Am Américium 241,06	96 Cm Curium 247,07	97 Bk Berkélium 249,08	98 Cf Californium 251,08	99 Es Einsteinium 254,09	100 Fm Fermium 257,10	101 Md Mendelevium 258,10	102 No Nobelium 255
89 Ac** Actinium 227,03	90 Th Thorium 232,04	91 Pa Protactinium 231,04	92 U Uranium 238,03	93 Np Neptunium 237,05	94 Pu Plutonium 239,05	95 Am Américium 241,06	96 Cm Curium 247,07	97 Bk Berkélium 249,08	98 Cf Californium 251,08	99 Es Einsteinium 254,09	100 Fm Fermium 257,10	101 Md Mendelevium 258,10	102 No Nobelium 255																		