

INTRODUCTION AU COUPLAGE CHROMATOGRAPHIE – SPECTROMETRIE DE MASSE

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Licence Pro
1er mars 2024
Université, Nantes





Le LABERCA

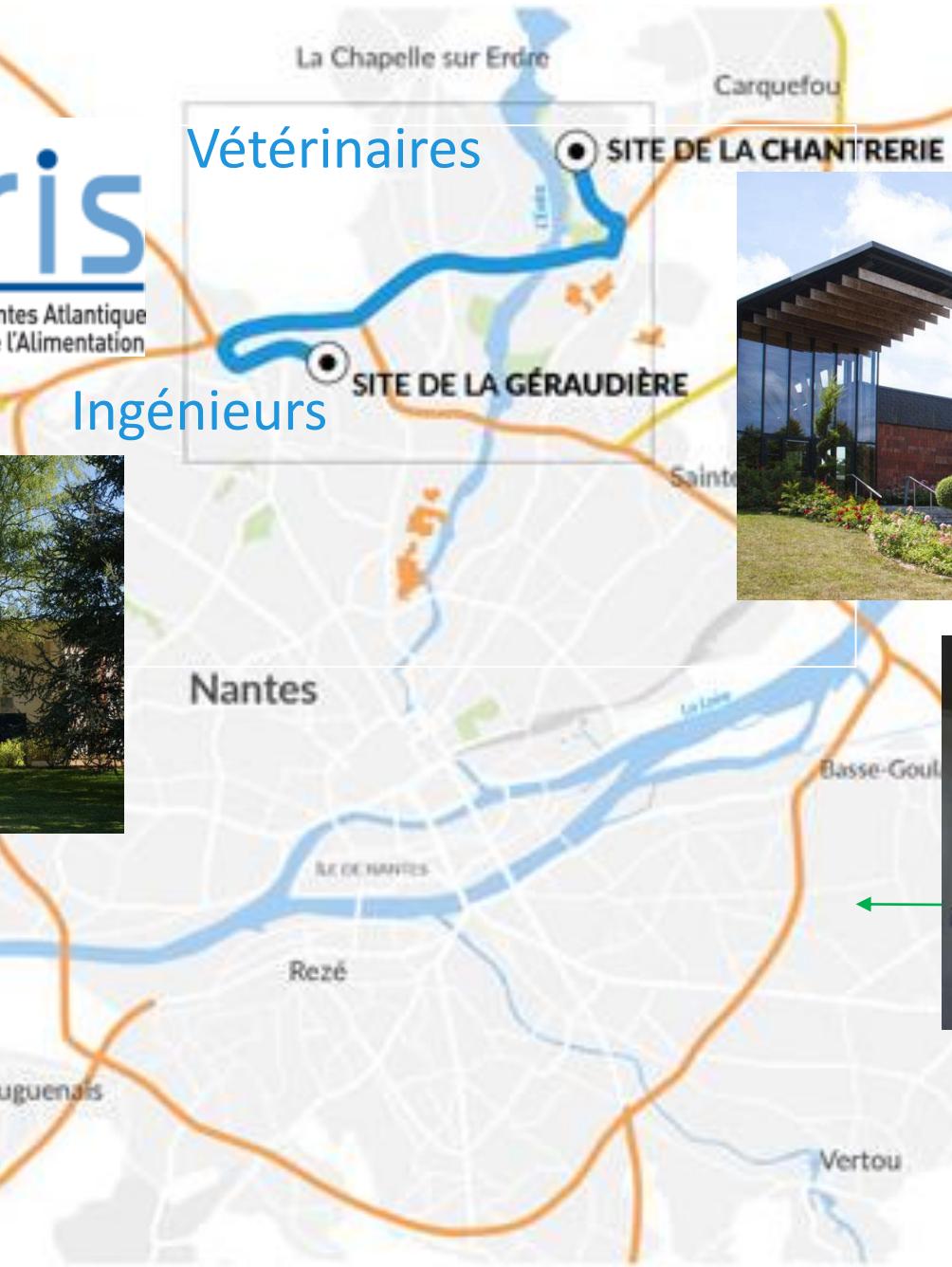


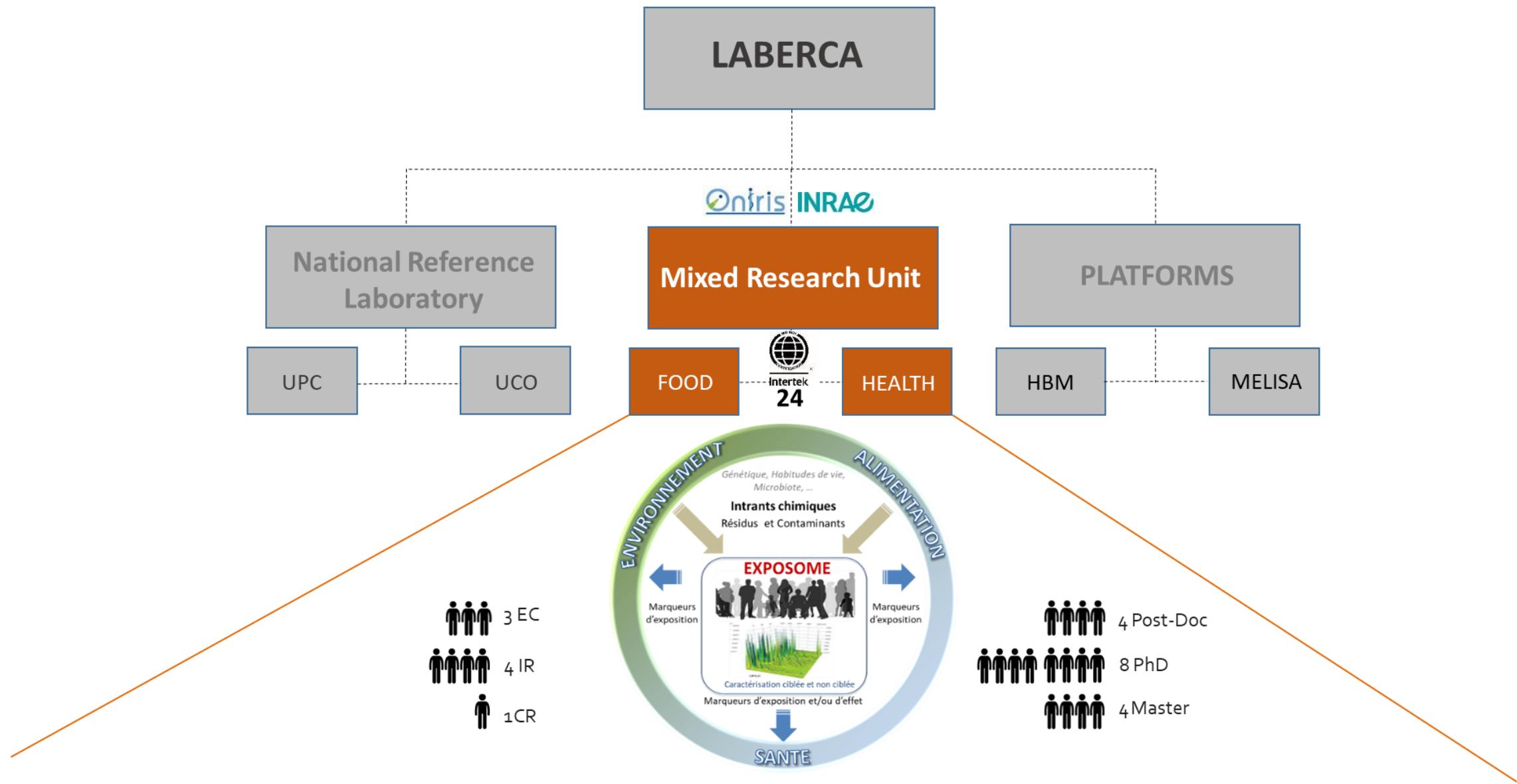
École Nationale
Vétérinaire, Agroalimentaire et de l'Alimentation

Nantes Atlantique



Ingénieurs





Global research thematic: organic chemical residues and contaminants from their environmental sources to their impact on human health through the food chain

ENVIRONMENT

Captation of Emerging Risks

- Detect and identify emerging chemical hazards
- Characterise their fate to the human food chain

FOOD

Chemical Food Safety

- Expand our knowledge of the human external chemical exposure
- Characterise the associated risk
- Guide and support the public policy with regard to risk assessment and risk management

TOXICOKINETIC MODELS

- Model data that are not available or easily accessible (e.g. due to cost, technical, ethical reasons)
- Link external and internal exposure
- Predict internal dose in given biological compartments
- Model lifespan exposure

ANIMAL MODEL

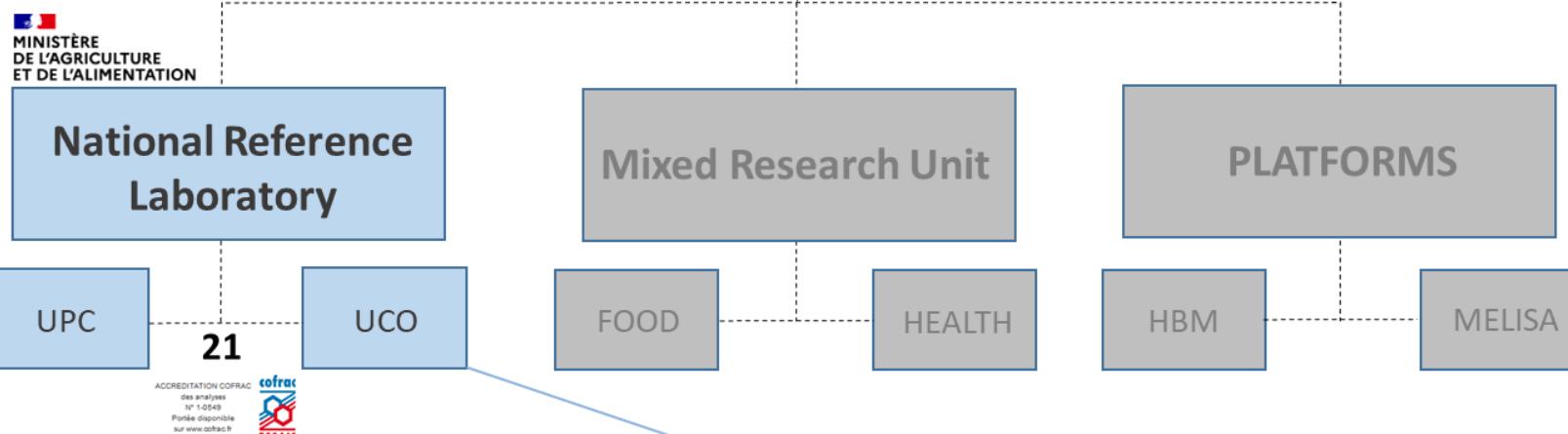
- Characterize the biological effects induced by chemical exposure
- Support human studies through exploratory and/or confirmatory study designs

HEALTH

Exposure – health relationships

- Expand our knowledge of the human internal chemical exposure
- Link internal exposure to environmental/food determinants
- Investigate the associated metabolic profile disruptions (effect markers)
- Investigate the link between exposure and health

LABERCA



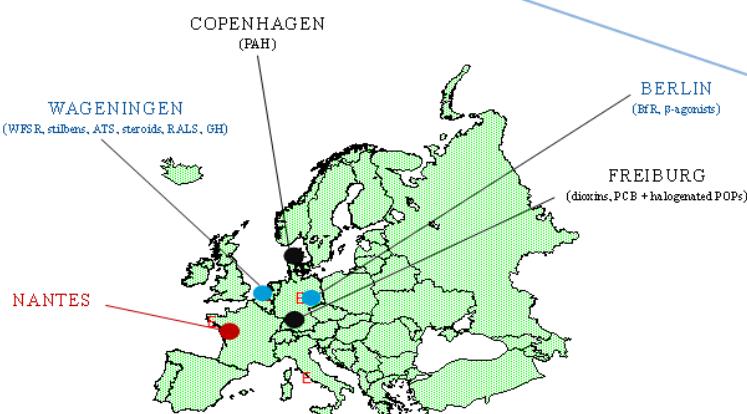
30 novembre 2018 JOURNAL OFFICIEL DE LA RÉPUBLIQUE FRANÇAISE Texte 82 sur 198

Décrets, arrêtés, circulaires

TEXTES GÉNÉRAUX

MINISTÈRE DE L'AGRICULTURE ET DE L'ALIMENTATION

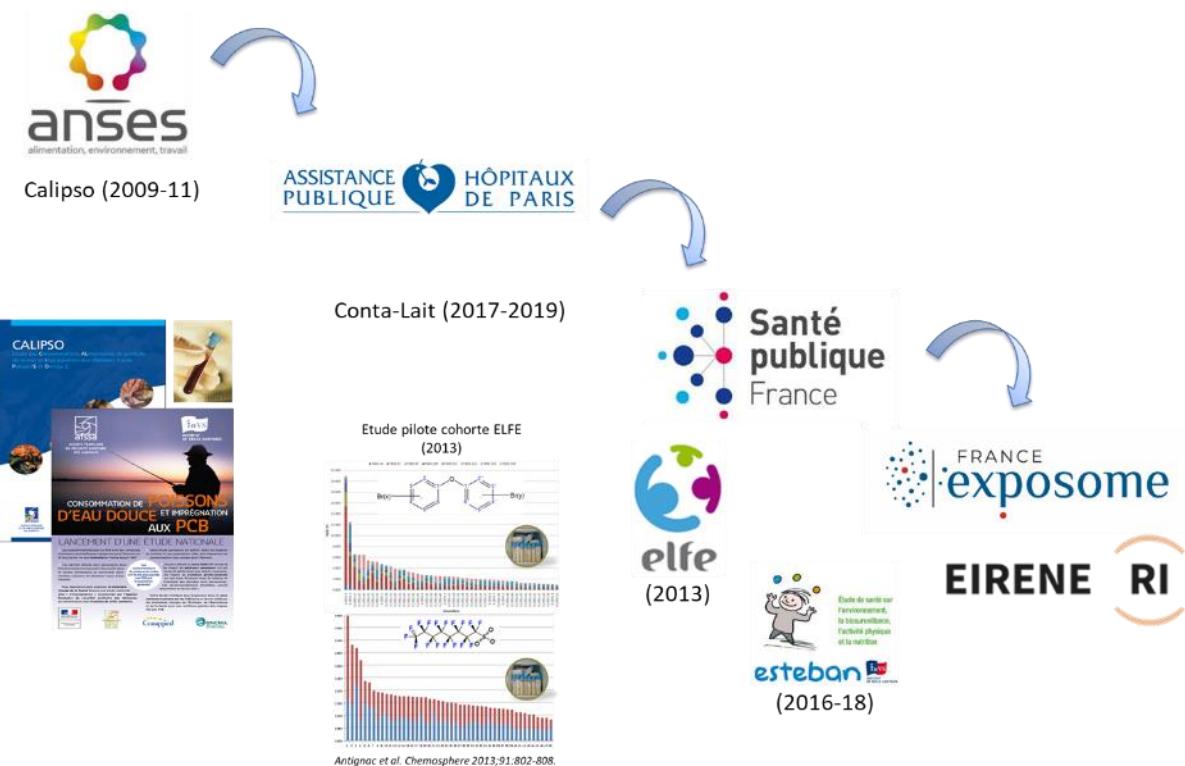
Arrêté du 26 novembre 2018 modifiant l'arrêté du 29 décembre 2009 désignant les laboratoires nationaux de référence dans le domaine de la santé publique vétérinaire et phytosanitaire

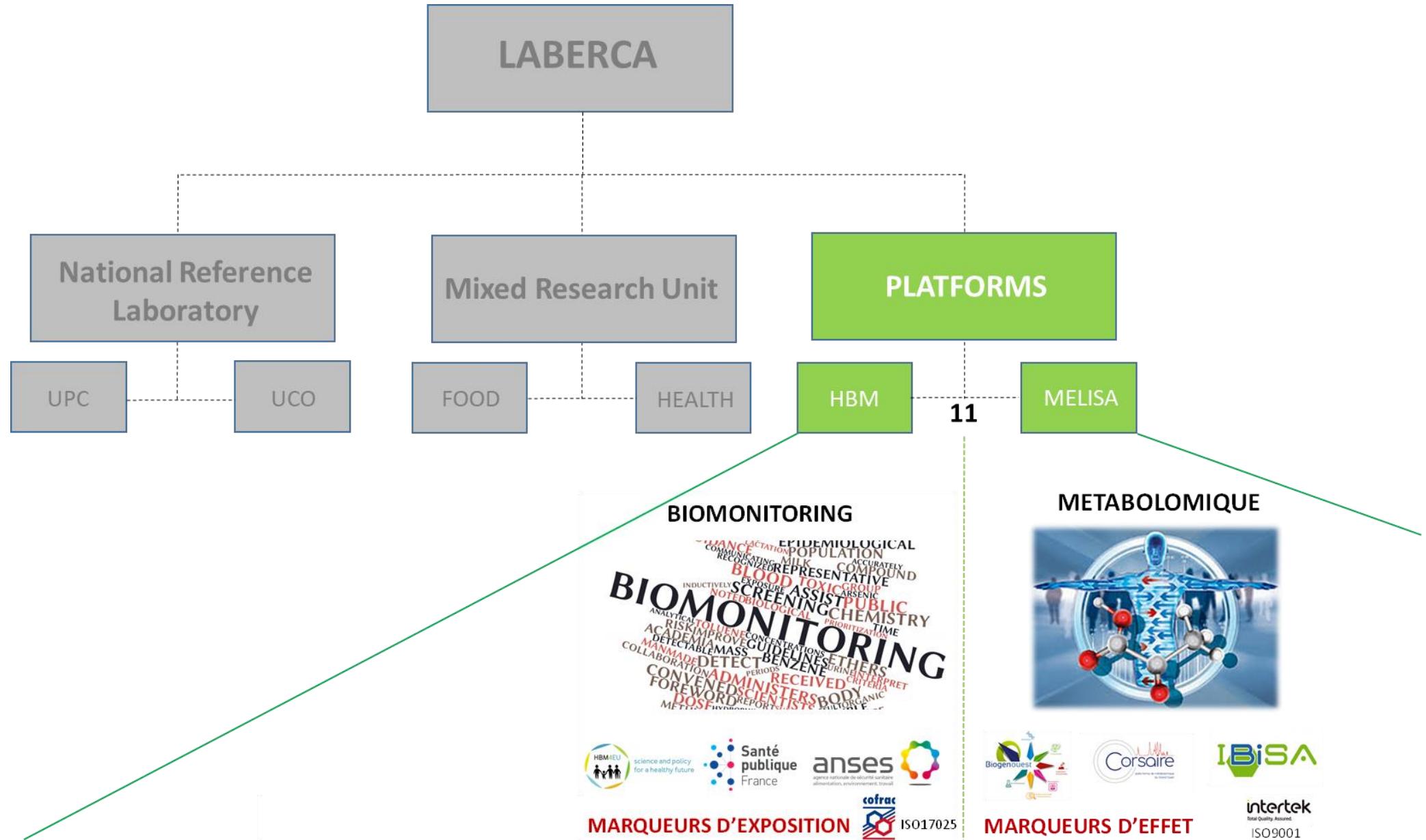


Characterizing the external (food) exposure e.g. total diet studies



Characterizing the internal exposure e.g. biomonitoring programs







LC-QTOF (IMS)



GCxGC-TOF



GC- and LC-HRMSⁿ (x 4, Orbitrap systems)



GC-HRMS (x 3, BE)



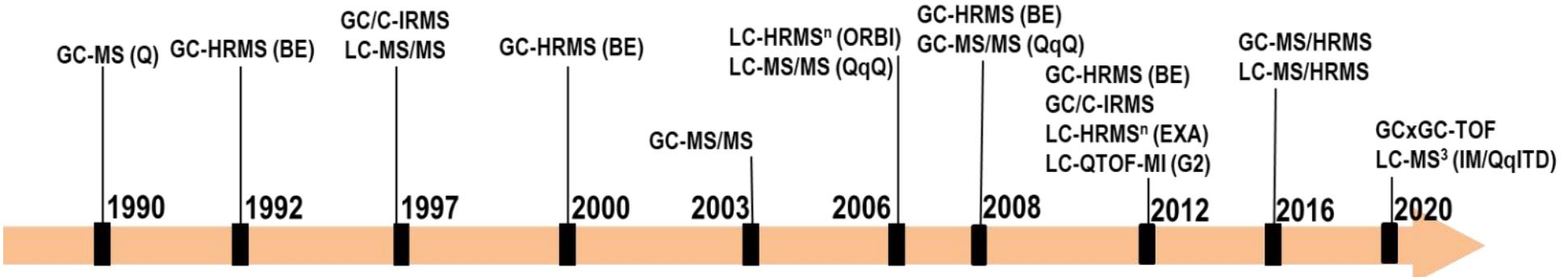
GC-C-IRMS (x 2)

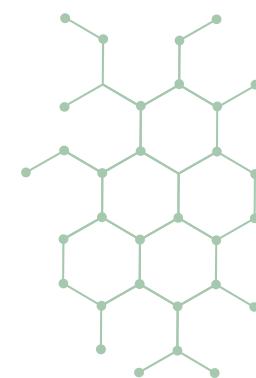
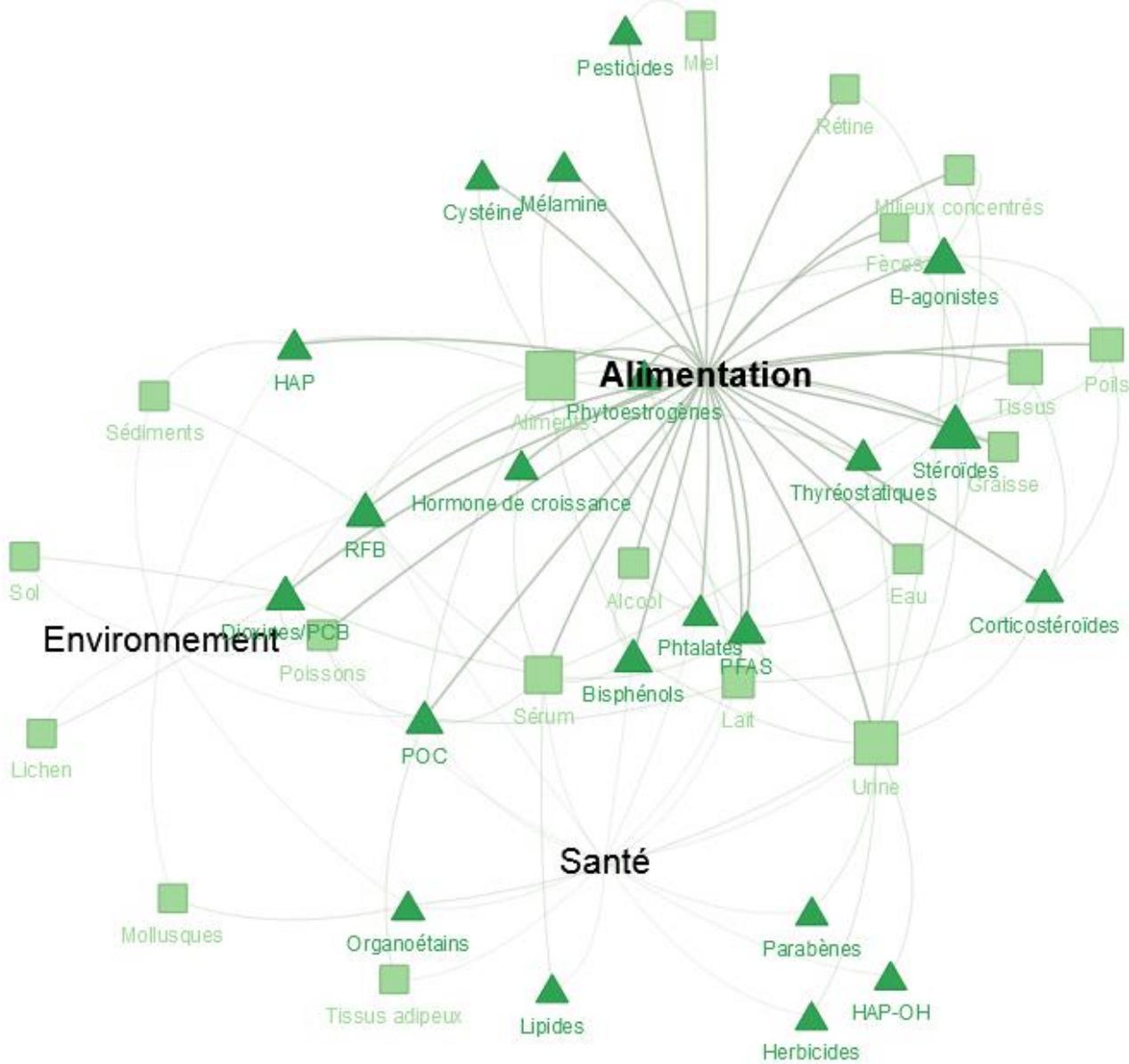


GC-MS/MS (x 3, QqQ)



LC-MS/MS (x 3, QqQ)





Analyte(s)



Source :
<http://visite.artsetmetiers.free.fr/instruments.html>

10^{-9} to 10^{-15} grams

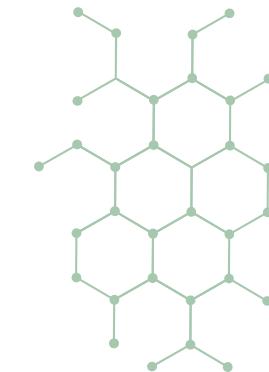


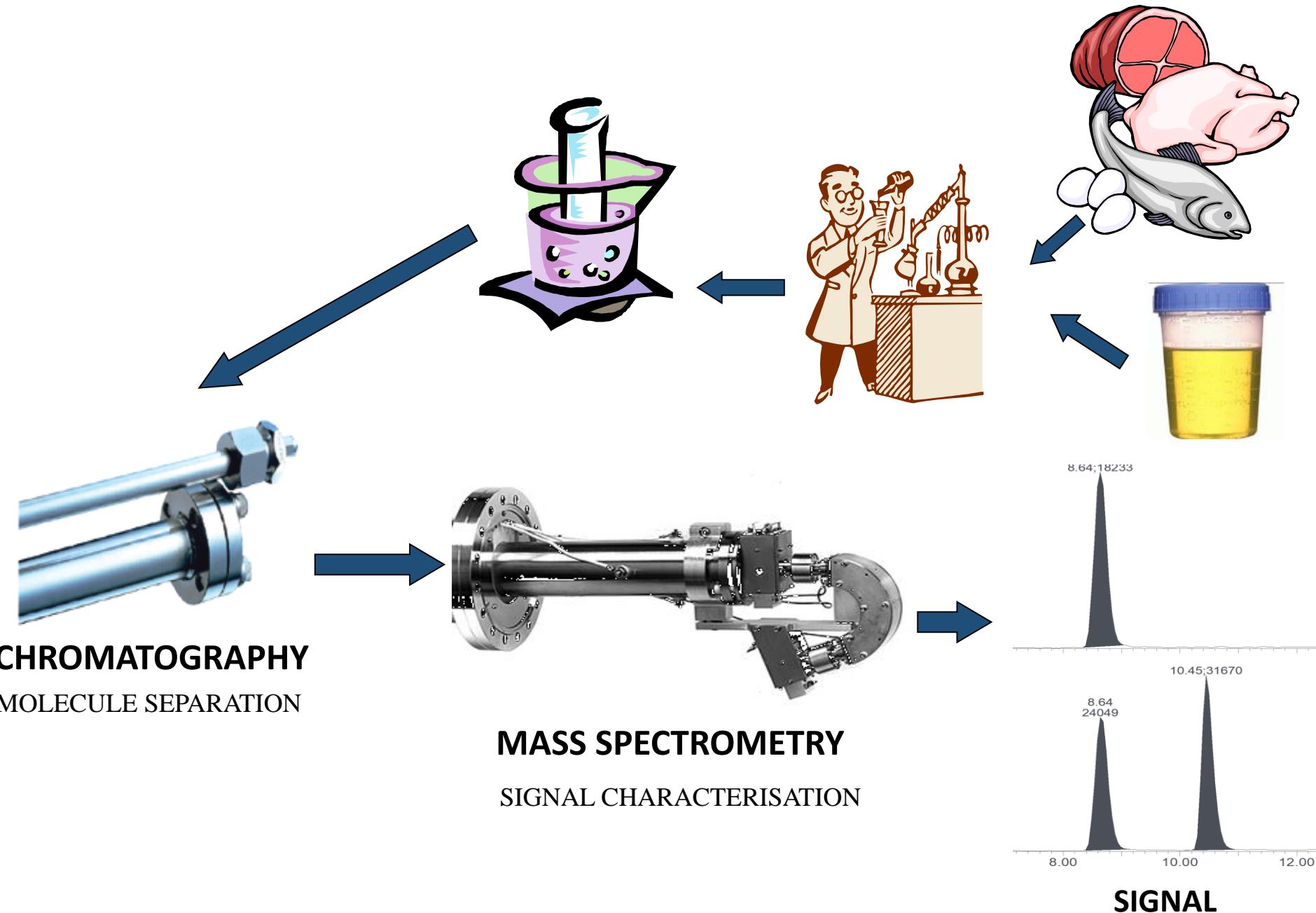
Matrix

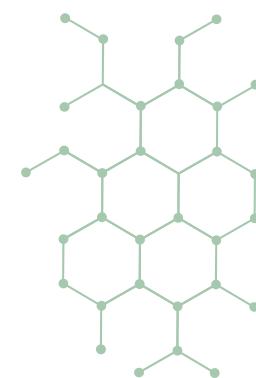
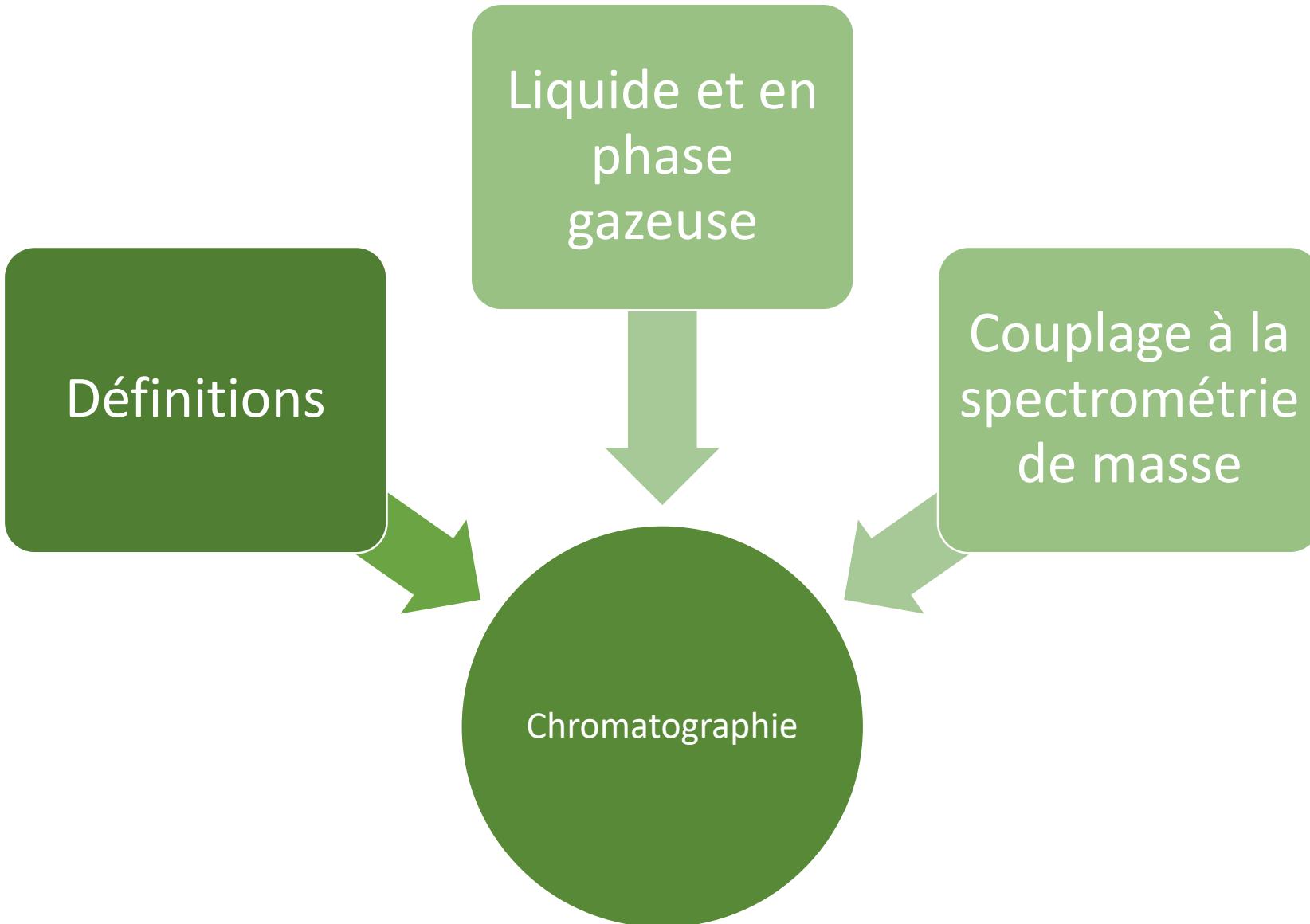


Source :
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grams



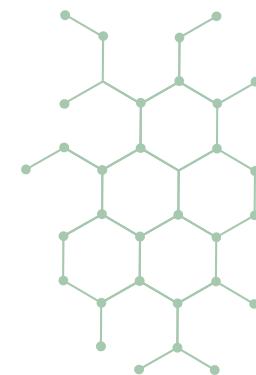




Définitions



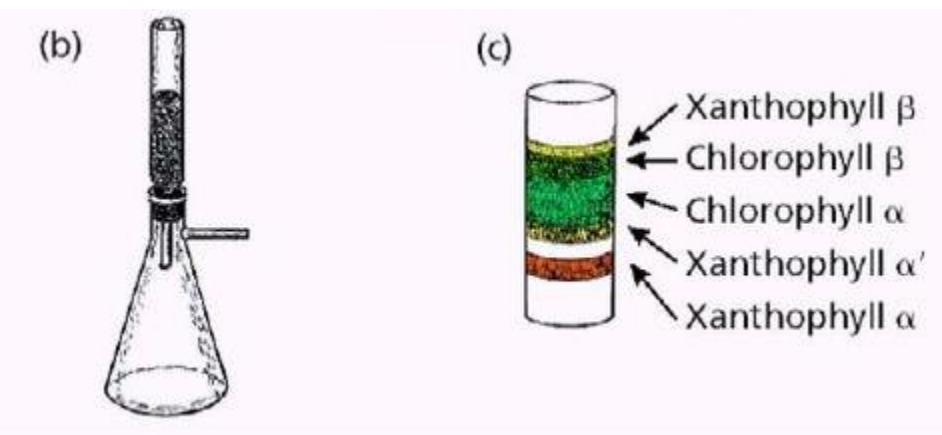
Chromatographie



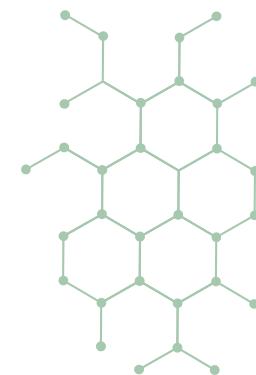
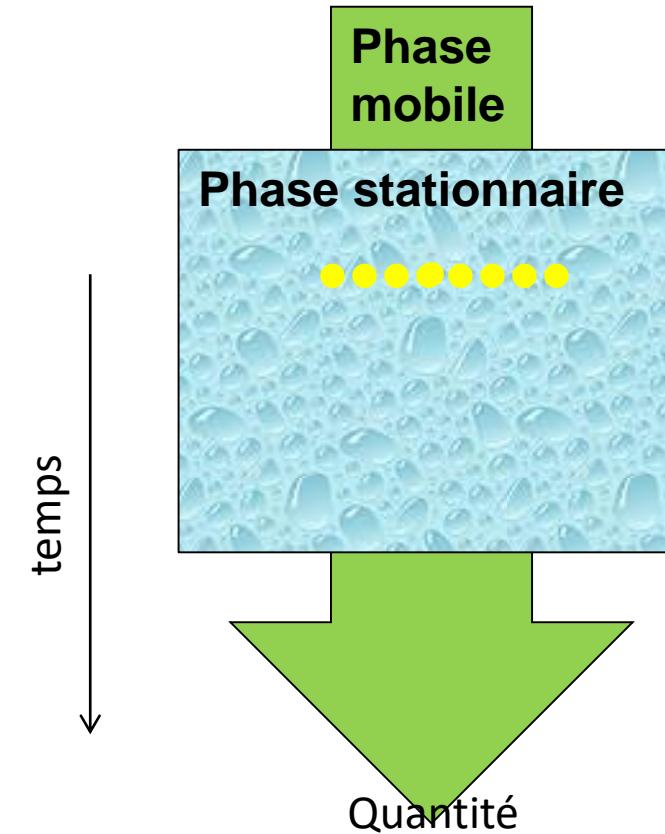
La chromatographie est une méthode séparative qui permet l'identification et le dosage des différents composés d'un mélange.

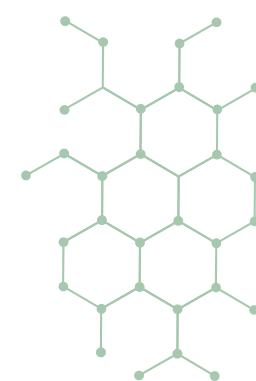
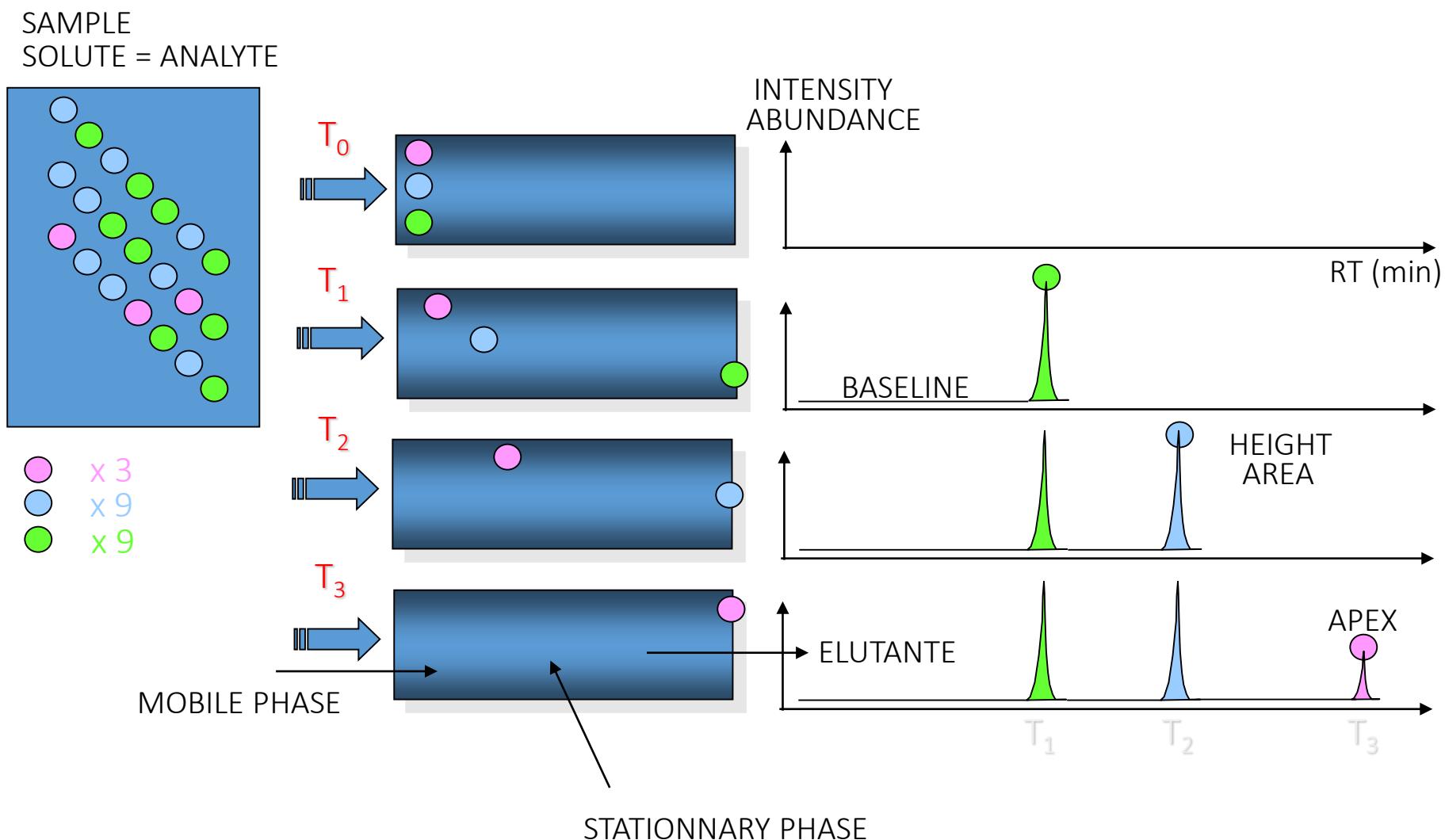
Michail Semenovich
Tswett – 1903

Séparation de pigments végétaux sur colonne remplie de carbonate de calcium
Il sépare la chlorophylle des caroténoïdes



Le principe est basé sur les différences d'affinité des composés du mélange avec la phase stationnaire et la phase mobile.





Type de chromatographie

- d'adsorption
- de partage
- d'échange d'ions
- d'exclusion

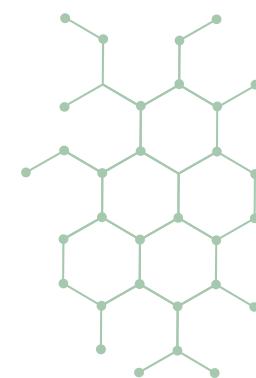
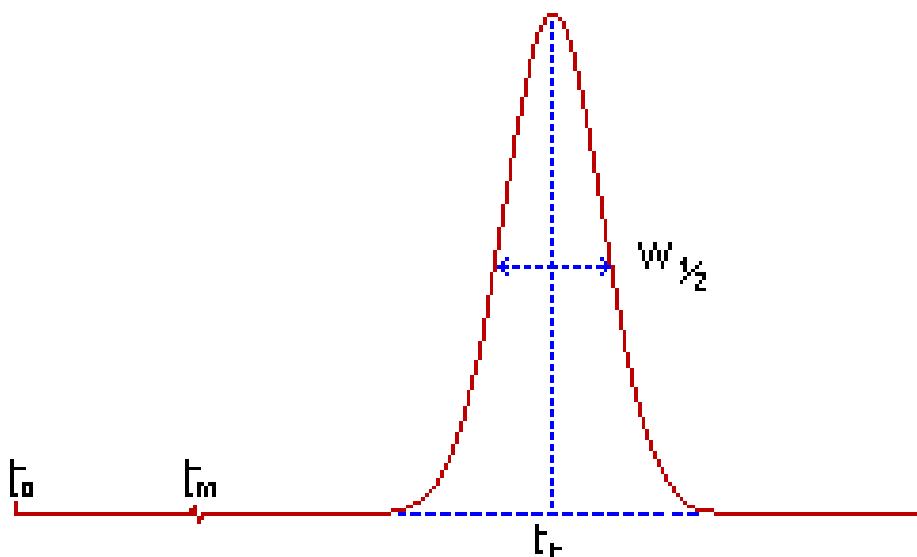
Phase mobile

- liquide
- gazeuse
- fluide supercritique

Volume mort (V_m au t_m)

Temps de rétention (t_r)

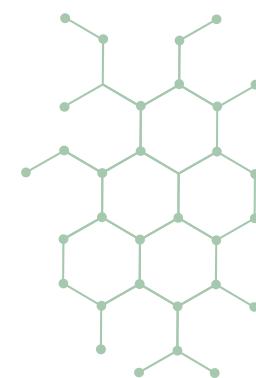
Capacité de rétention $k = (t_r - t_m) / t_m$



Chromatographie
liquide



Chromatographie



Chromatographie Liquide

HPLC



MS



<http://autosamplervials.kinesis.co.uk/products/vials-caps/lx-lgc-certified-vial-cap-kits-10#.VO3utPmG-FU>

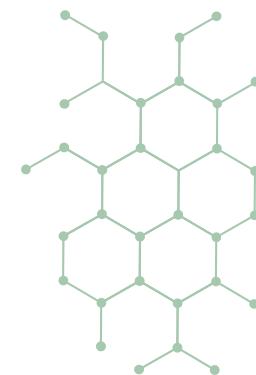
Passeur d'échantillon

Injecteur

Four pour colonne

Pompe HPLC

MS



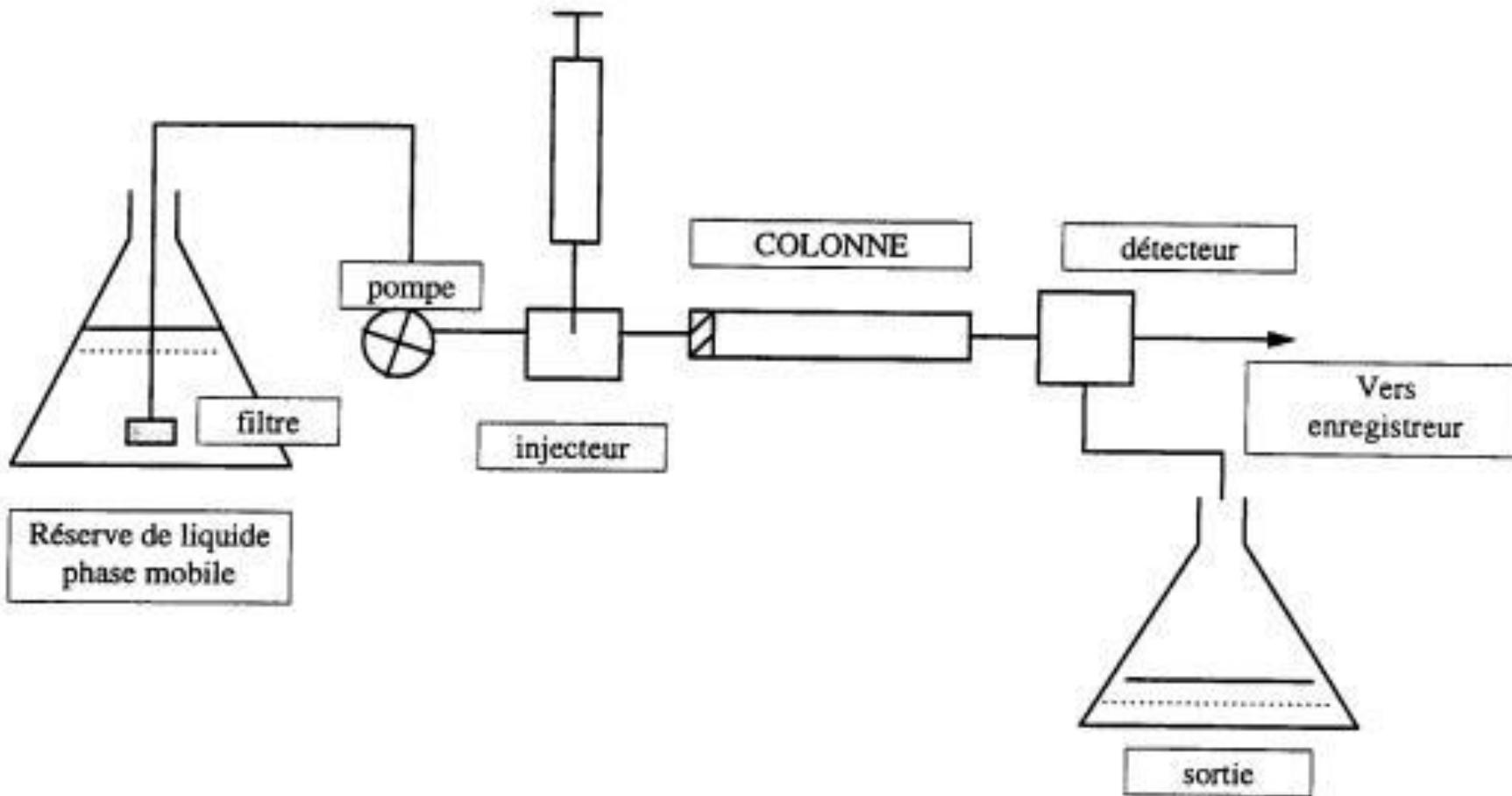
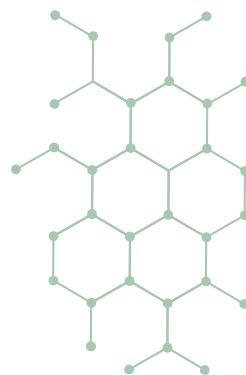


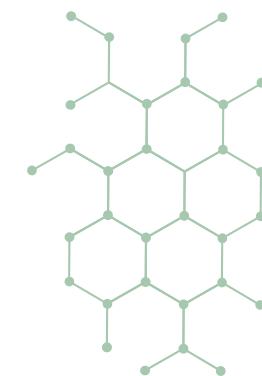
Figure 3 : principe de fonctionnement de l'HPLC

<http://www.chimie-sup.fr/chromatographie.htm>



HPLC Modes

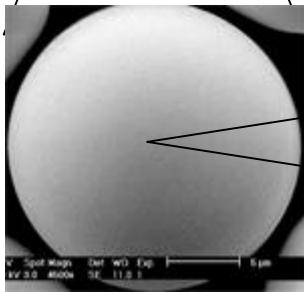
Analytes	Mode	Stationary phase	Mobile phase
Neutrals Weak acids Weak bases	Reversed-phase	C18, C8, Phenyl ... (Apolar)	Water/organic, modified
Ionics Acids Bases	Ion-pairing	C18, C8, Phenyl (Apolar)	Water/organic, ion-pairing reagent
Compounds insoluble in water Organic isomers	Normal phase	Silica, Diol, Amino, Cyano (Polar)	Organic
Ionics	Ion exchange	Ion exchange resin	Aqueous buffer
High MW compounds Polymers	Size exclusion	Polymeric or silica	Aqueous – Gel filtration Organic – Gel permeation
Polar	HILIC	Silica	Organic (ACN) / water



Chromatographie Liquide

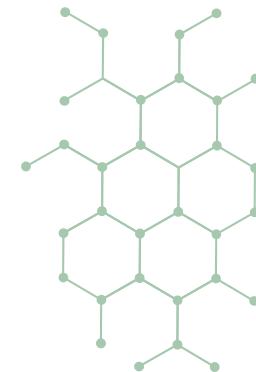
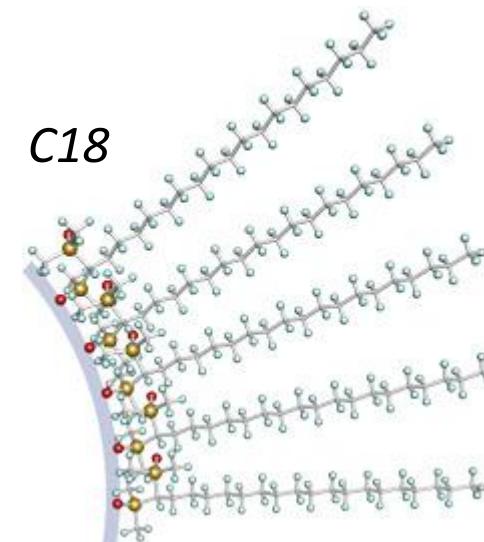
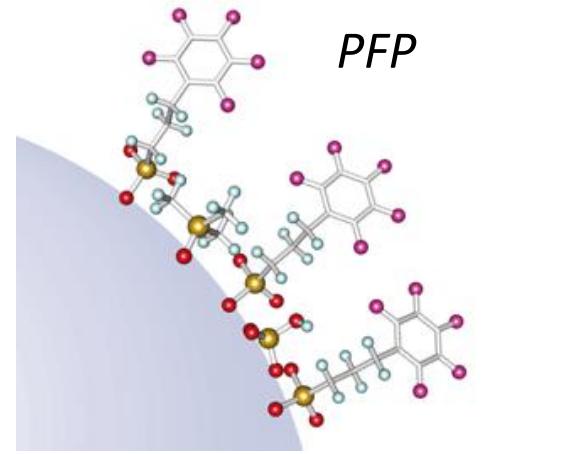


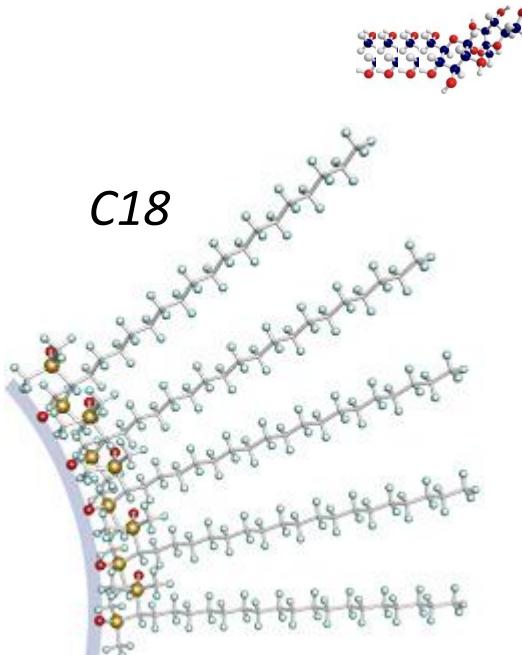
100 x 2 mm



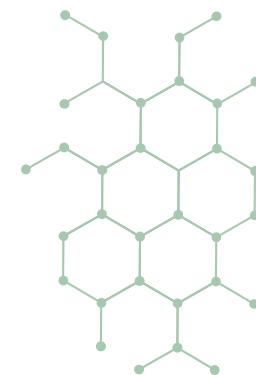
↔

2 μm





**2-augmentation du
pouvoir éluant**



Chromatographie Liquide

Force éluante et polarité

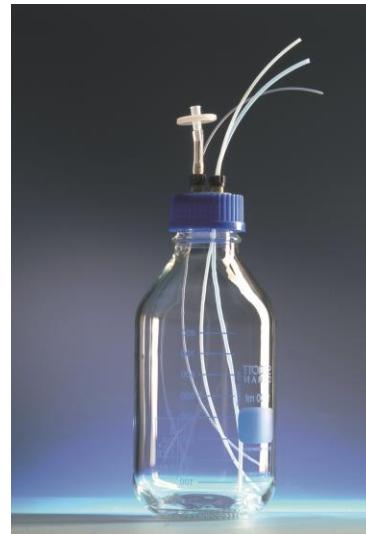
*Pour les composés **polaires** :*

- + la phase mobile sera **polaire**, + elle va entraîner les solutés.
- + la phase mobile sera **apolaire**, - elle va entraîner les solutés.

*Pour les composés **peu polaires** :*

- + la phase mobile sera **polaire**, - elle va entraîner les solutés
- + la phase mobile sera **apolaire**, + elle va entraîner les solutés.

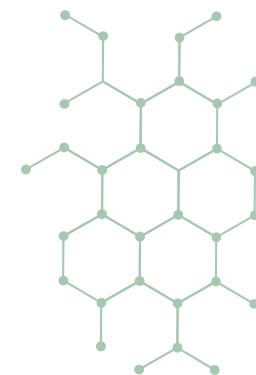
Solvant A
Phase aqueuse
Tampon



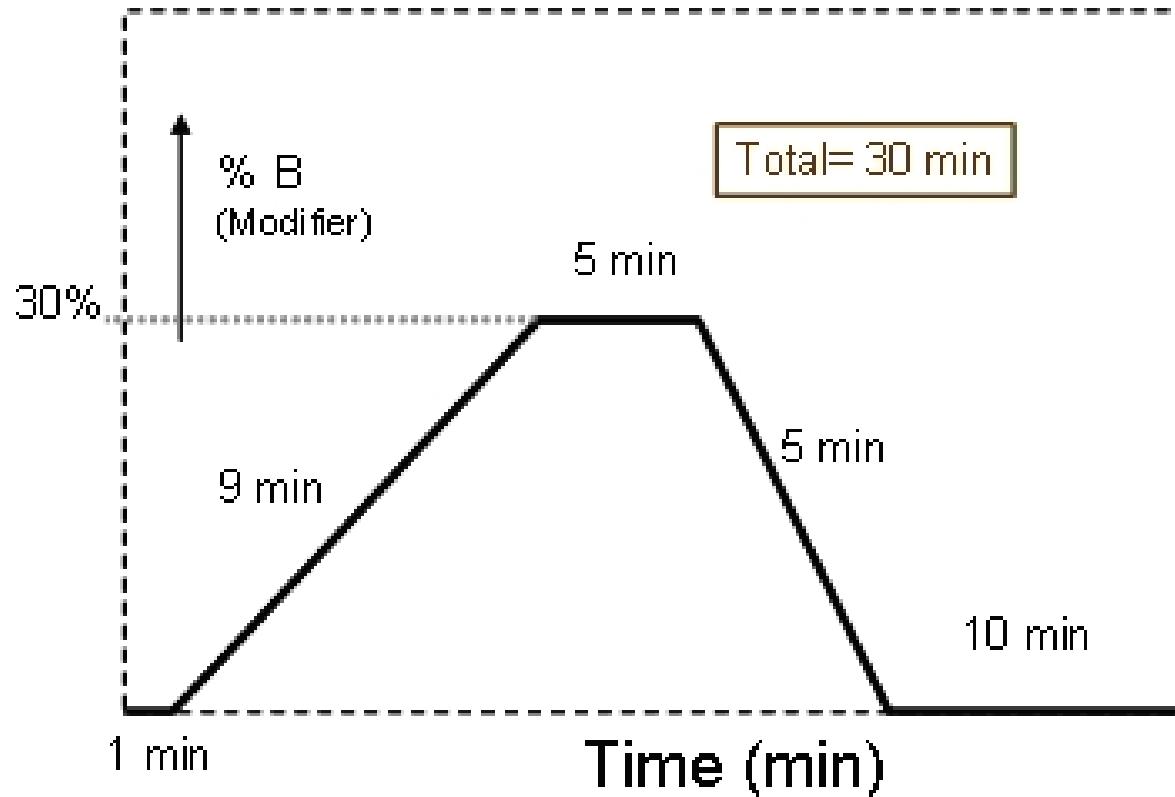
Solvant B
ACN
MeOH



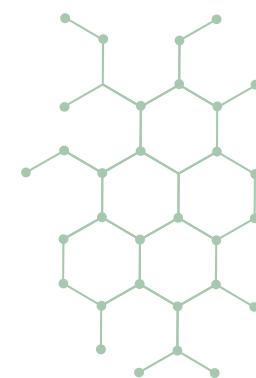
Vitesse (vitesse) importante pour la séparation : dépend de la colonne



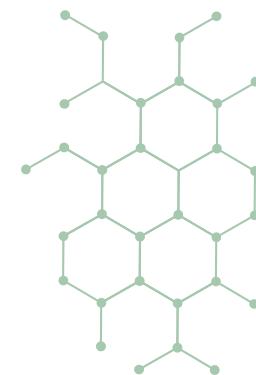
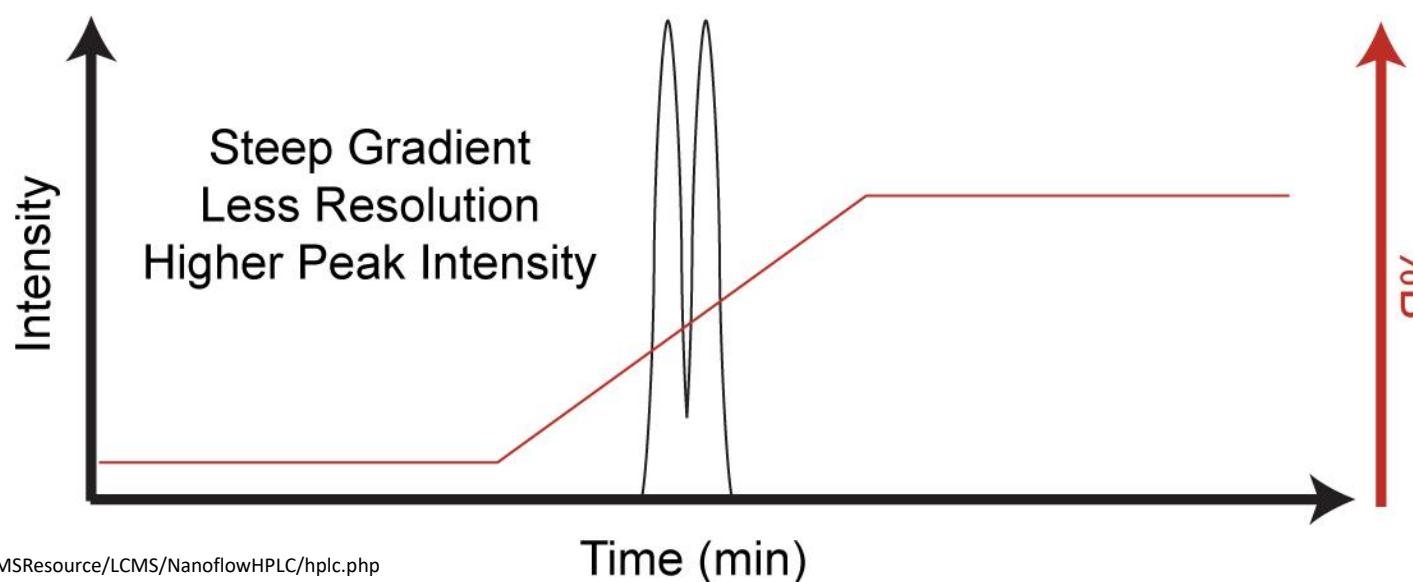
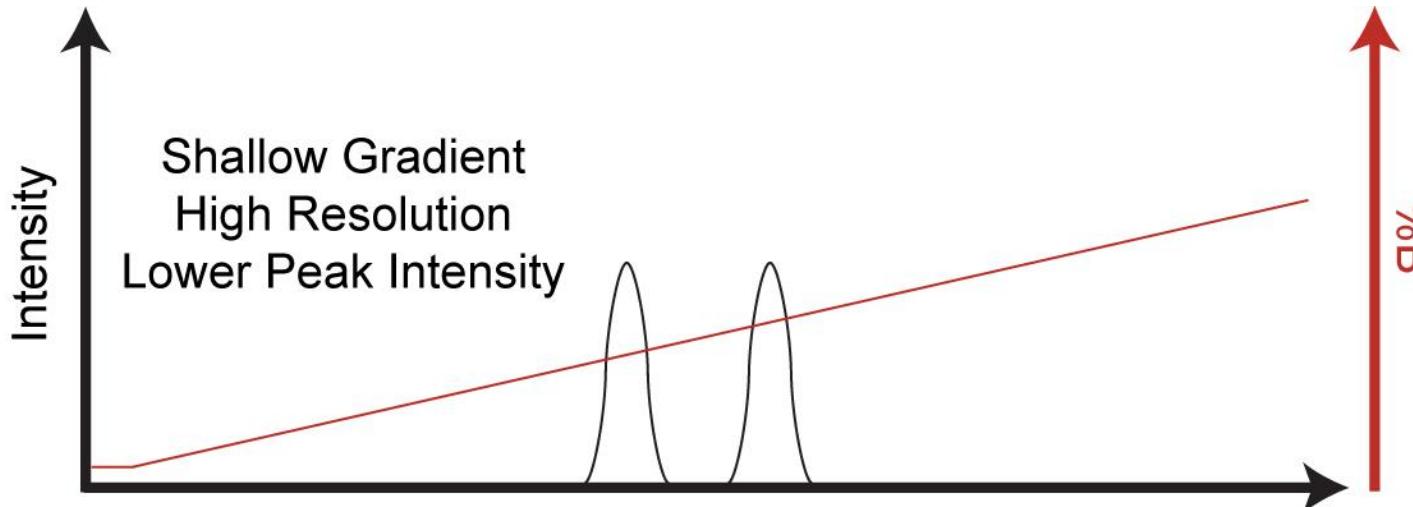
Gradient et isocratique



http://www.forumsci.co.il/HPLC/7_Isocratic_Gradient.html

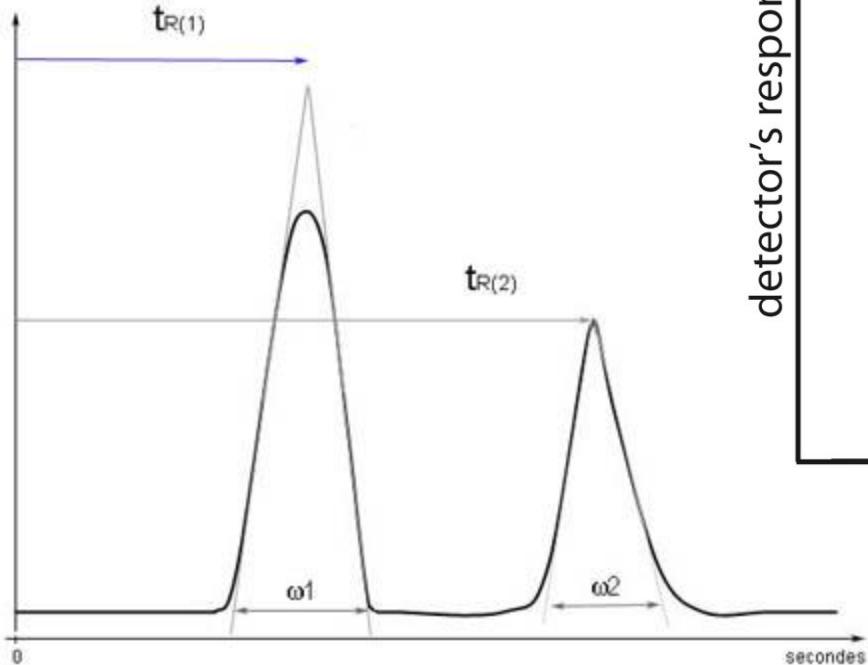


Chromatographie Liquide

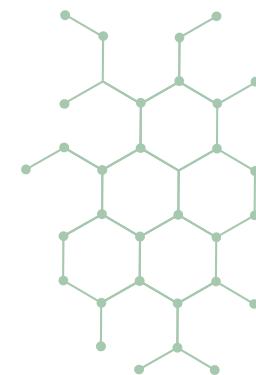
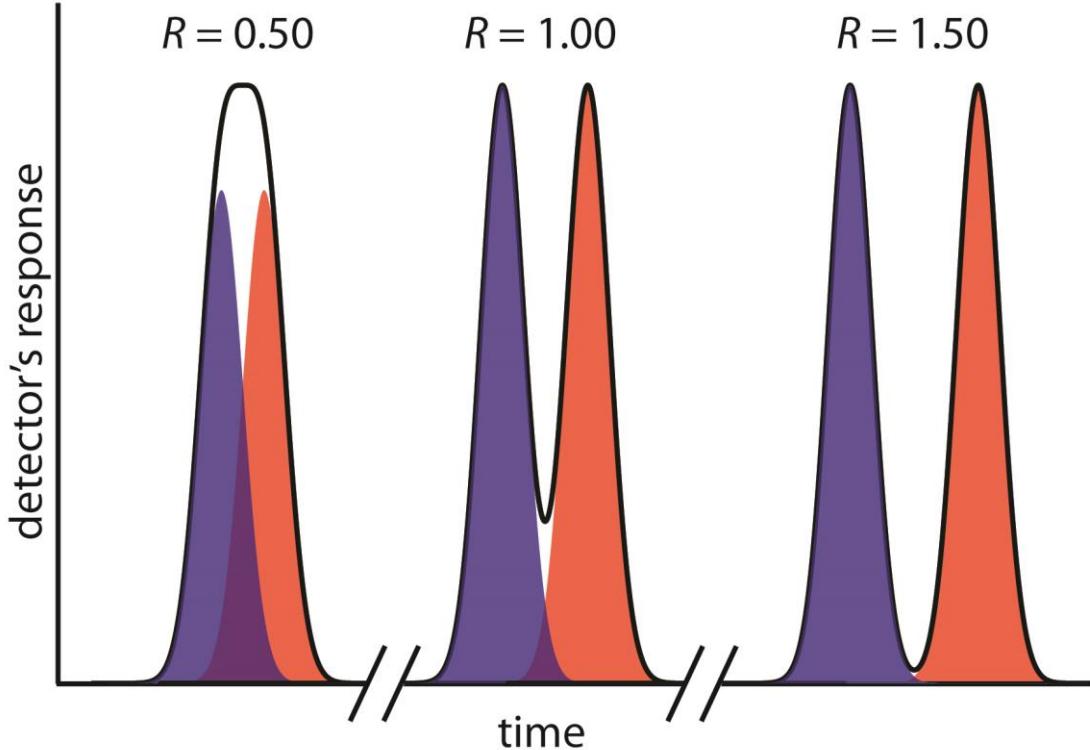


Résolution chromatographique

$$R = 2 \frac{(t_{r2} - t_{r1})}{(\omega_1 + \omega_2)}$$



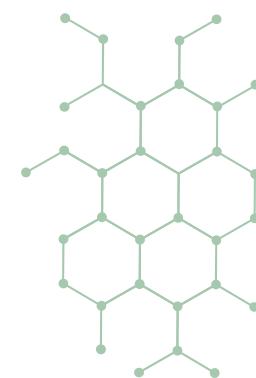
<http://www.lachimie.fr/analytique/chromatographie/facteur-resolution.php>



Chromatographie
en phase
gazeuse



Chromatographie



Chromatographie en Phase Gazeuse



Passeur
d'échantillon

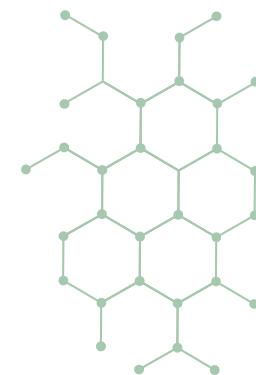
Injecteur

Four
pour
colonne

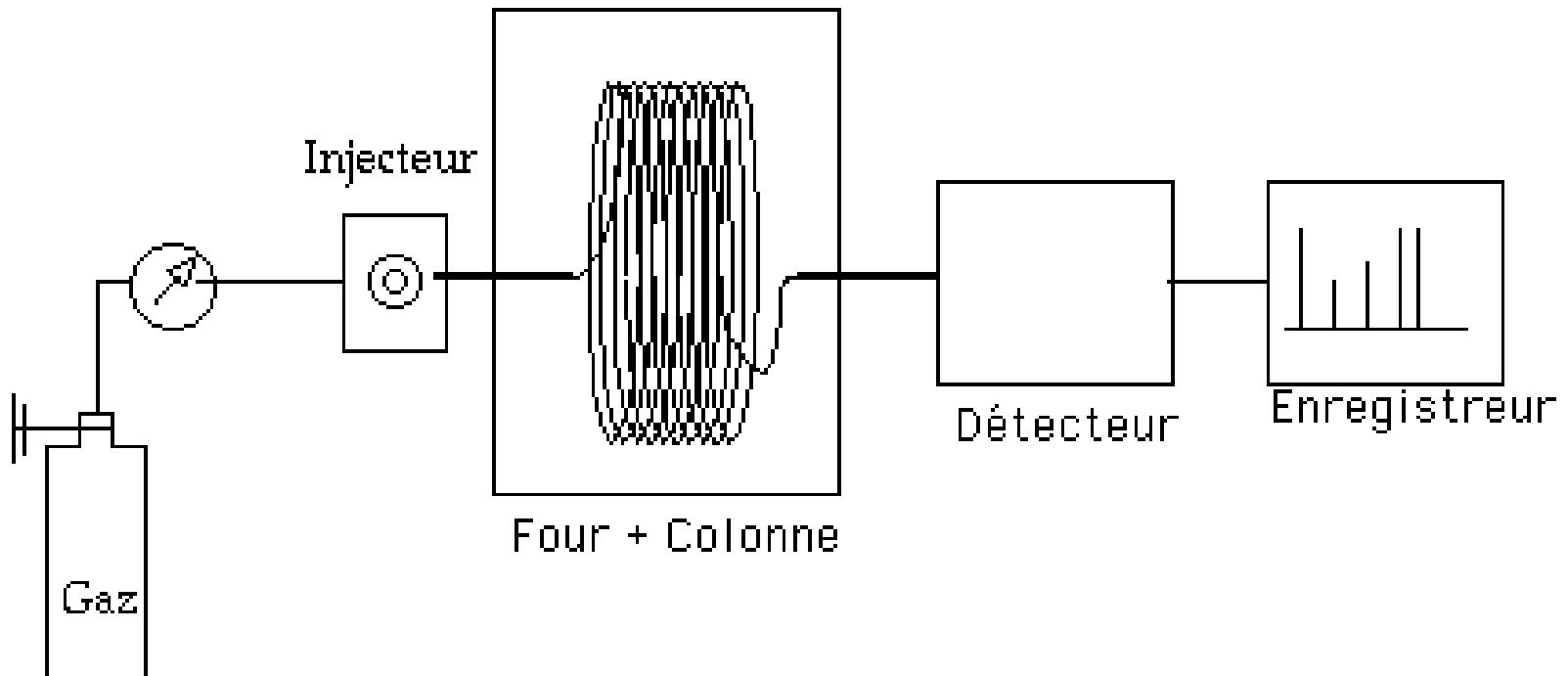
Ligne de
transfert

MS

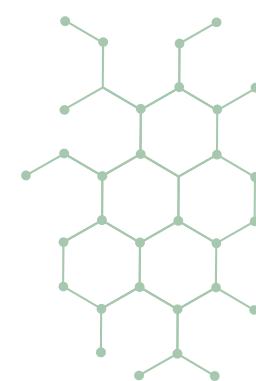
<http://autosamplervials.kinesis.co.uk/products/vials-caps/kx-lcgc-certified-vial-cap-kits-10#.VO3utPmG-FU>



Chromatographie en Phase Gazeuse

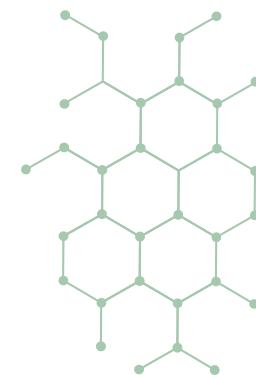
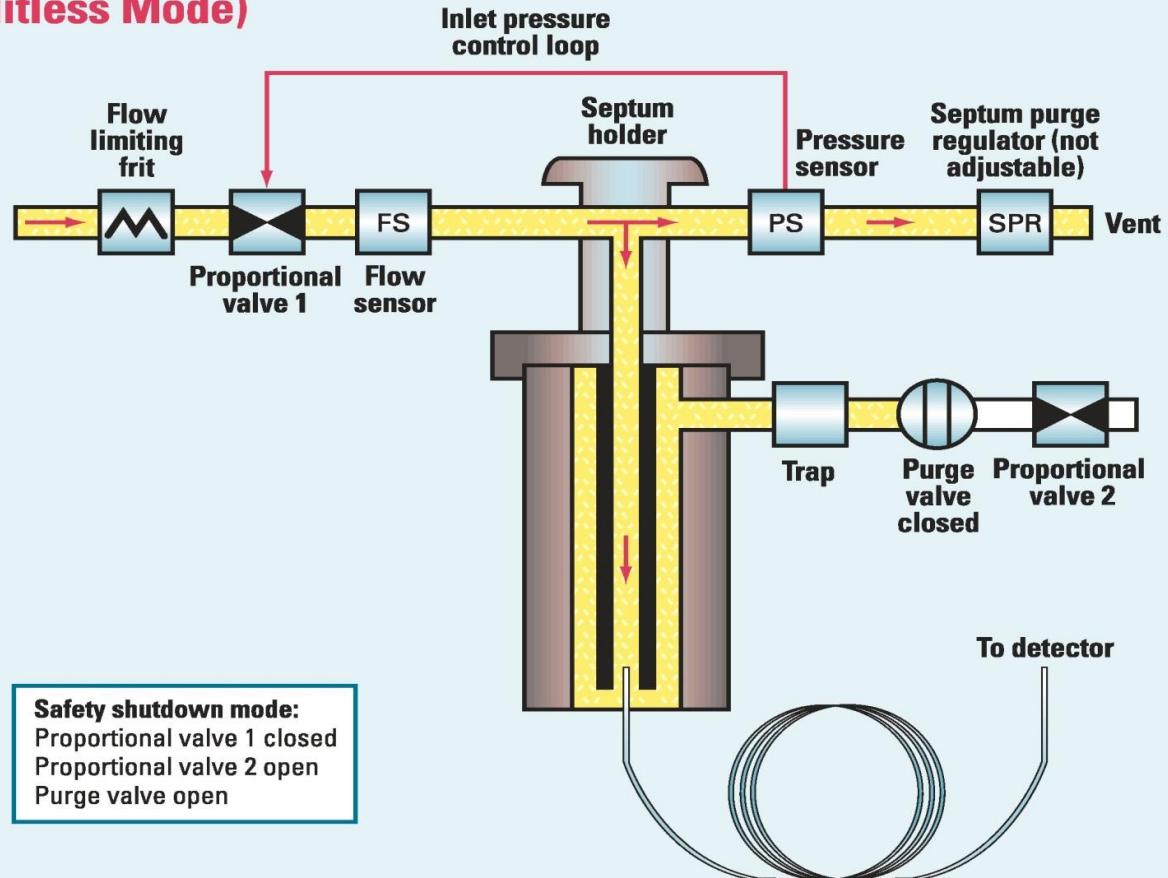


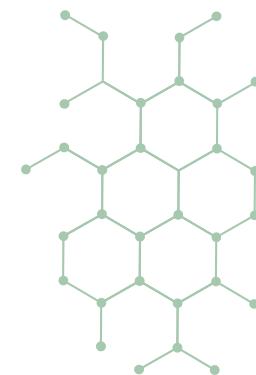
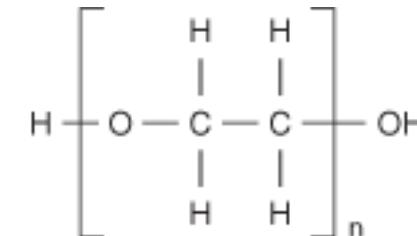
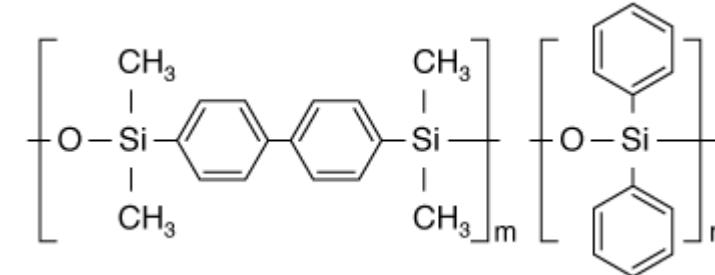
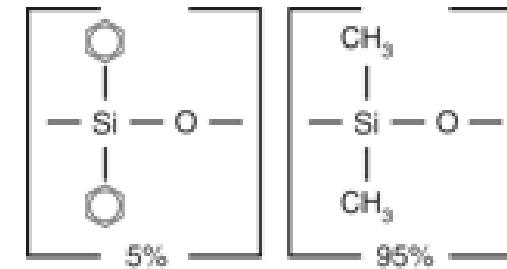
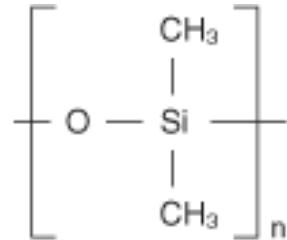
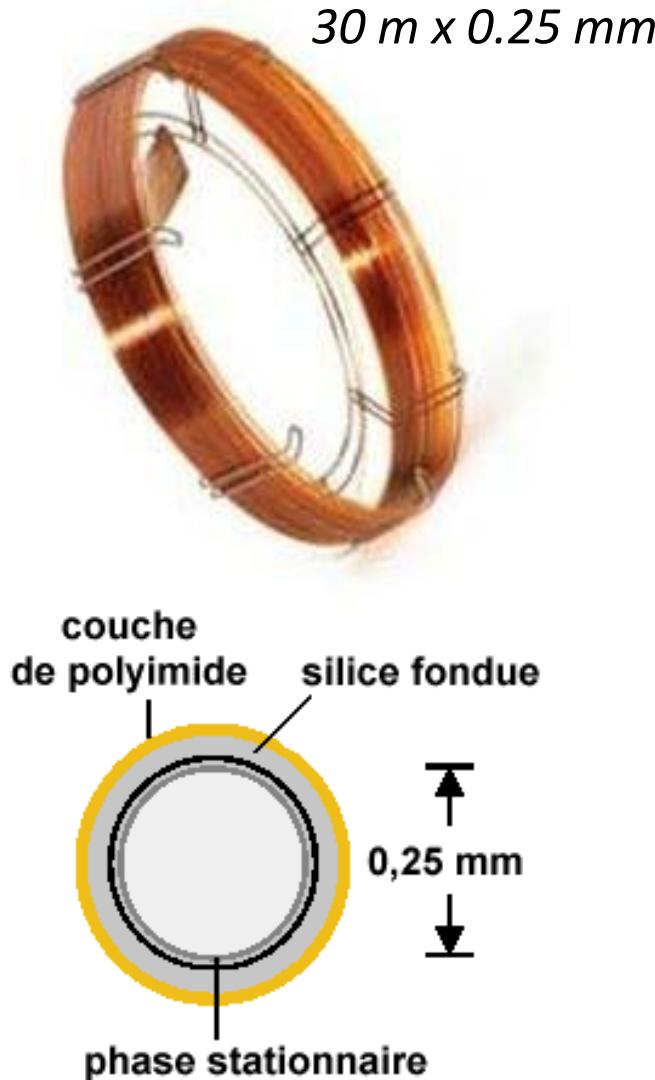
<http://www.snv.jussieu.fr/bmedia/lafont/chromato/A3.html>



6890/7890 Splitless Injector

Split/Splitless Inlet (Splitless Mode)





Etapes

1 - Injection avec le passeur automatique

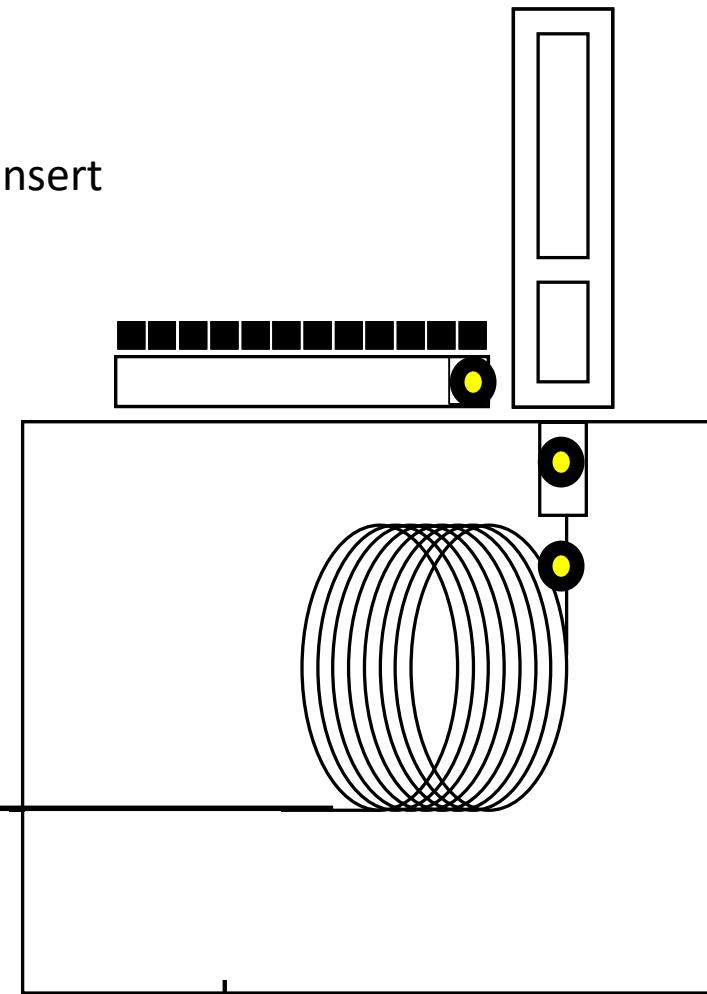
Injecteur à **250 °C**, expansion du solvant dans l'insert

2 – Dépôt de l'échantillon

Injecteur à **250 °C** non splitté pendant 1 min

Four à **90 °C** (Toluène)

Condensation en tête de colonne



3 – Séparation

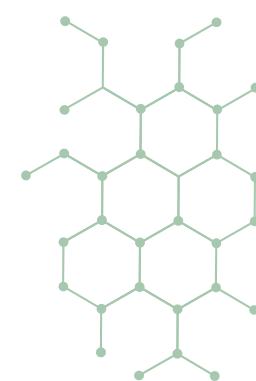
Injecteur à **250 °C** en purge après 1 min

Four : $T^\circ \nearrow$: TR toluene

4 – Elution

Injecteur à **250 °C** en purge après 1 min

Four : $T^\circ \nearrow$: TR Analyte

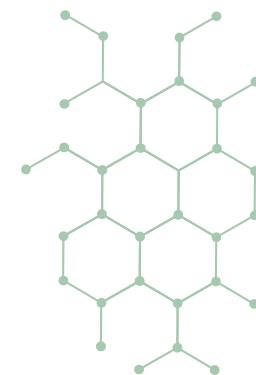
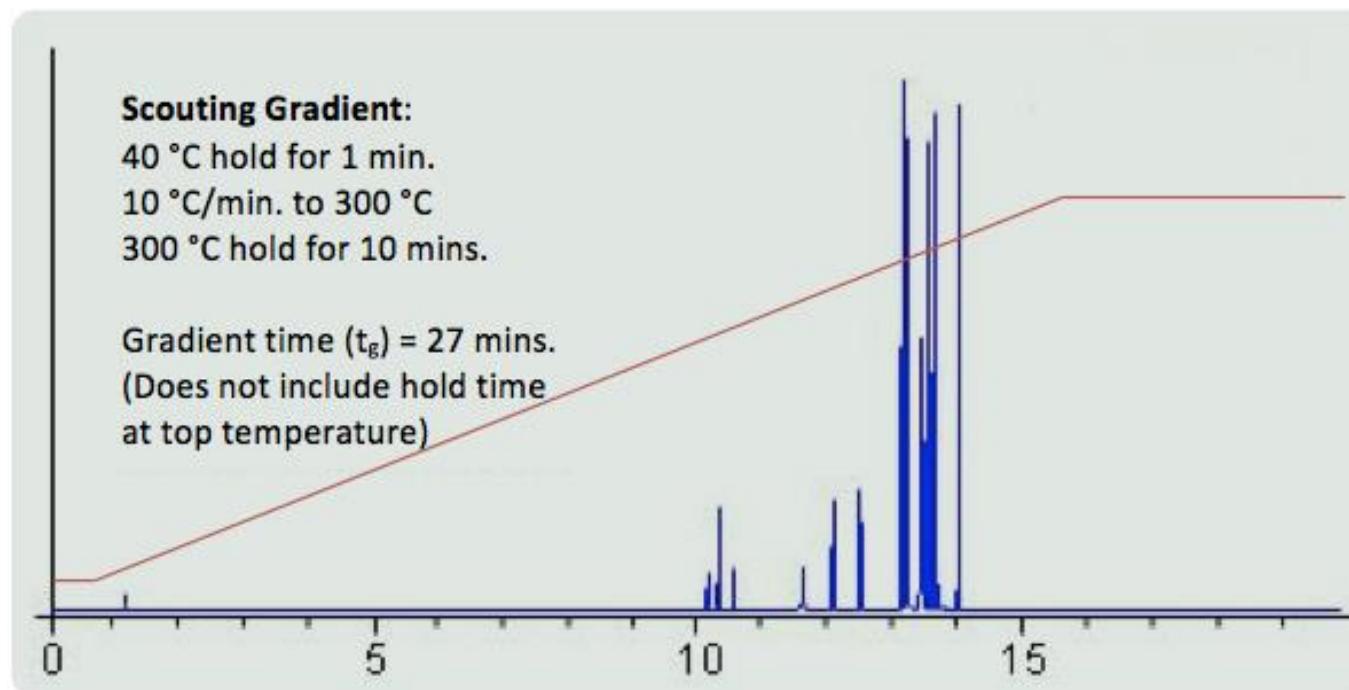


Elution

Gaz vecteur : Helium, H_2 voire N_2

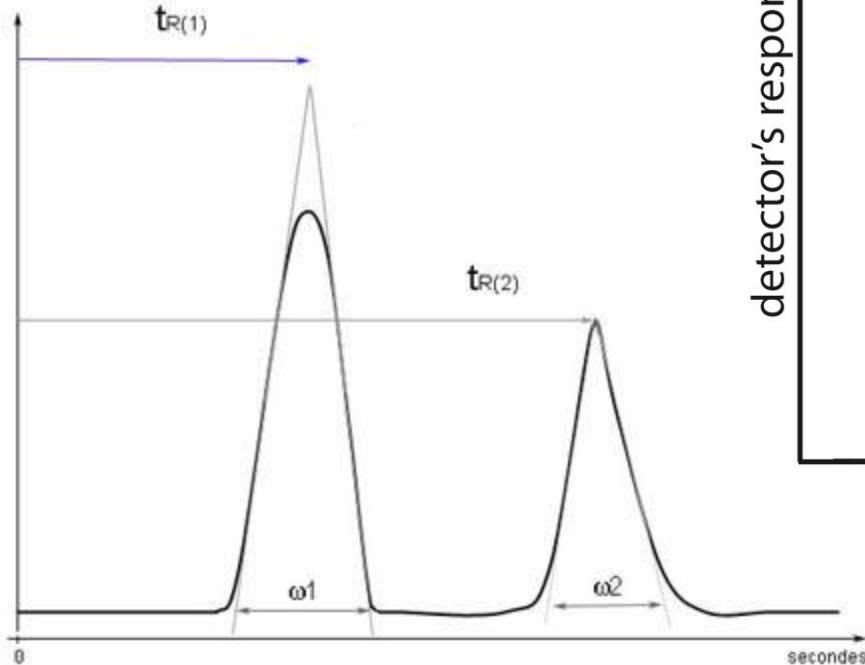
Vitesse (vitesse) importante pour la séparation : dépend de la colonne

Gradient et isotherme

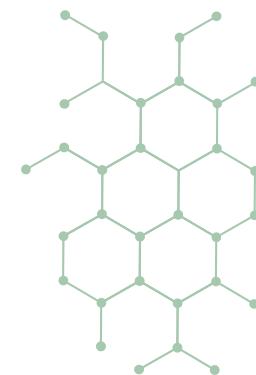
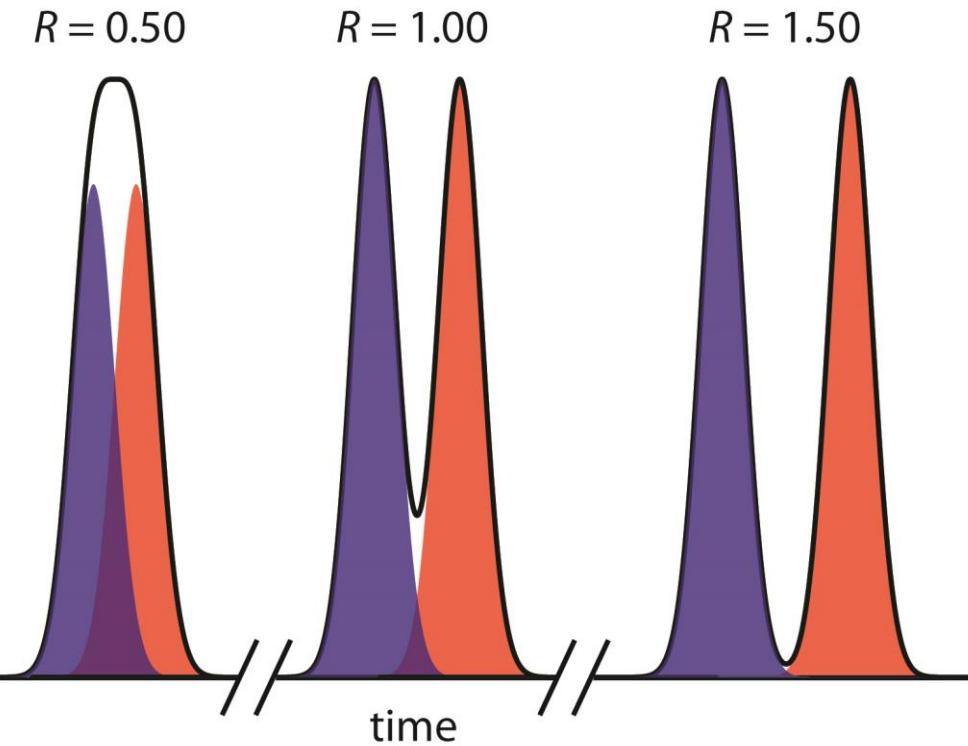


Résolution chromatographique

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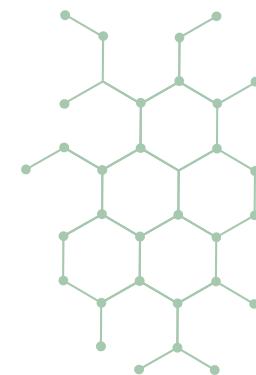
<http://www.lachimie.fr/analytique/chromatographie/facteur-resolution.php>



Introduction à la chromatographie

HPLC	GC	SFC
<ul style="list-style-type: none">• Organic molecules, biomolecules, ions, polymers• High pressure pumps• UV/VIS (Photodiode array (PDA) and Mass spectrometry based	<ul style="list-style-type: none">• Organic Inorganic Must be volatile• Carrier Gas (i.e Helium, Nitrogen• FID ,TCD, MS, ECD, FPD, PID	<ul style="list-style-type: none">• Organic molecules• Mobile phase = CO₂ + modifier• High pressure pump + ABPR• Same detectors as LC

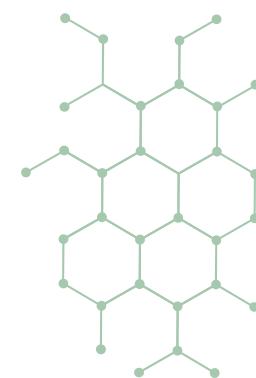
Dans des conditions identiques (colonne, température, débit), TR répétable



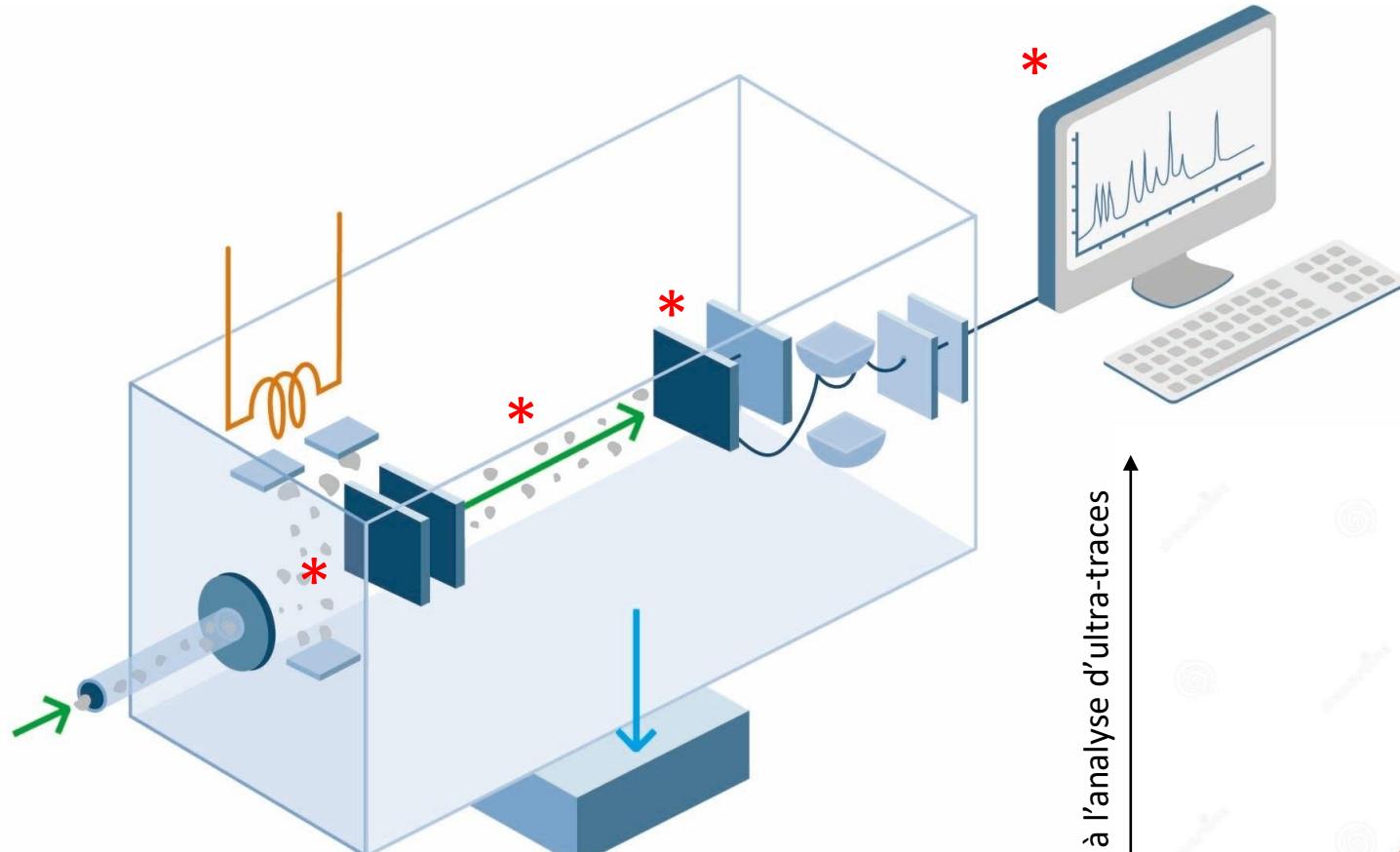
Couplage à la
spectrométrie
de masse



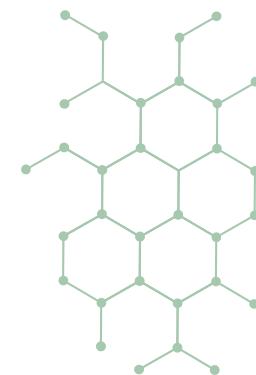
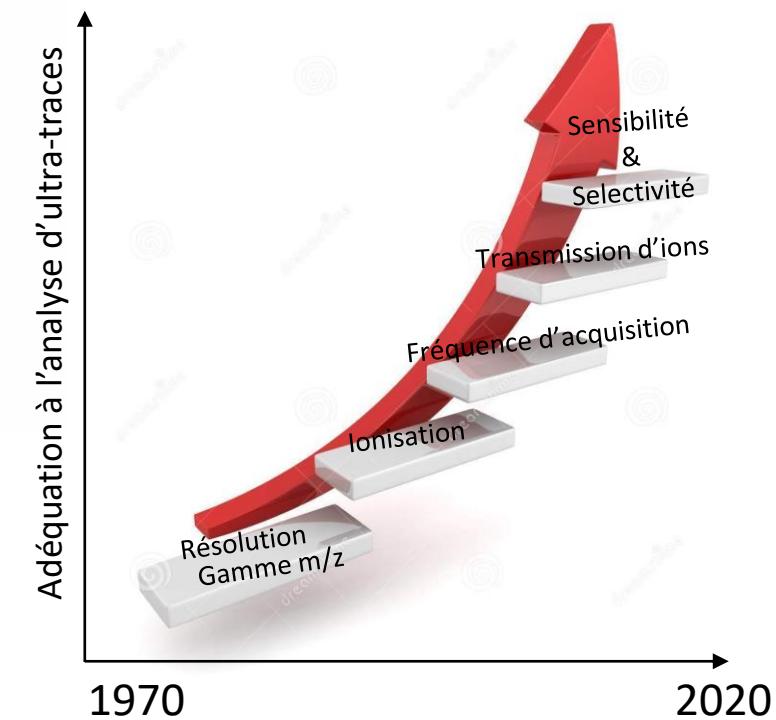
Chromatographie



Spectrométrie de masse

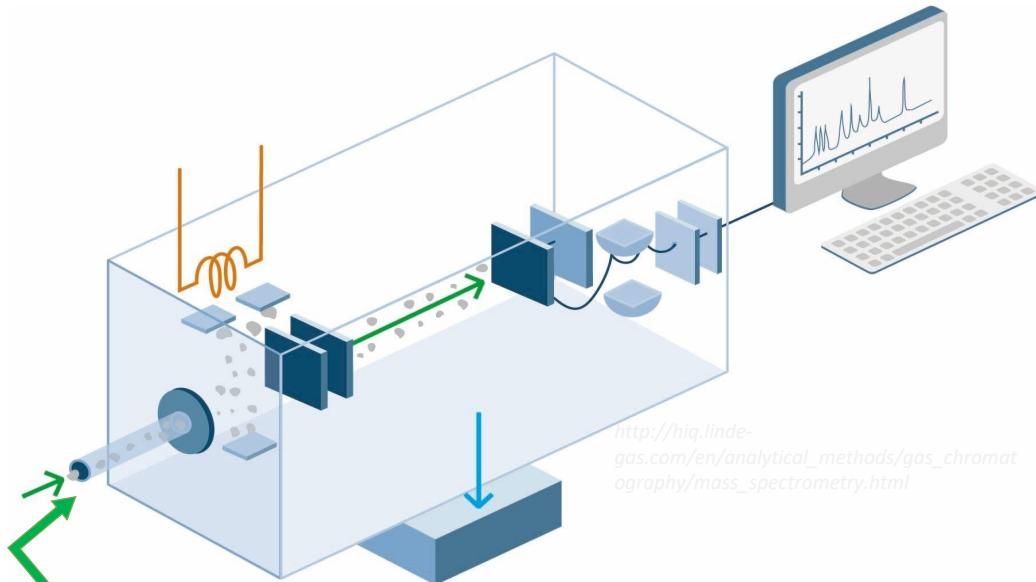


http://hiq.linde-gas.com/en/analytical_methods/gas_chromatography/mass_spectrometry.html



Spectrométrie de masse

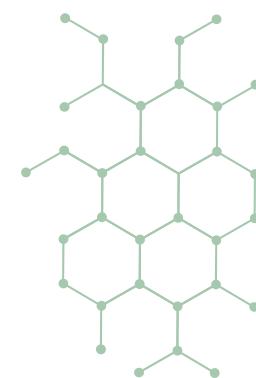
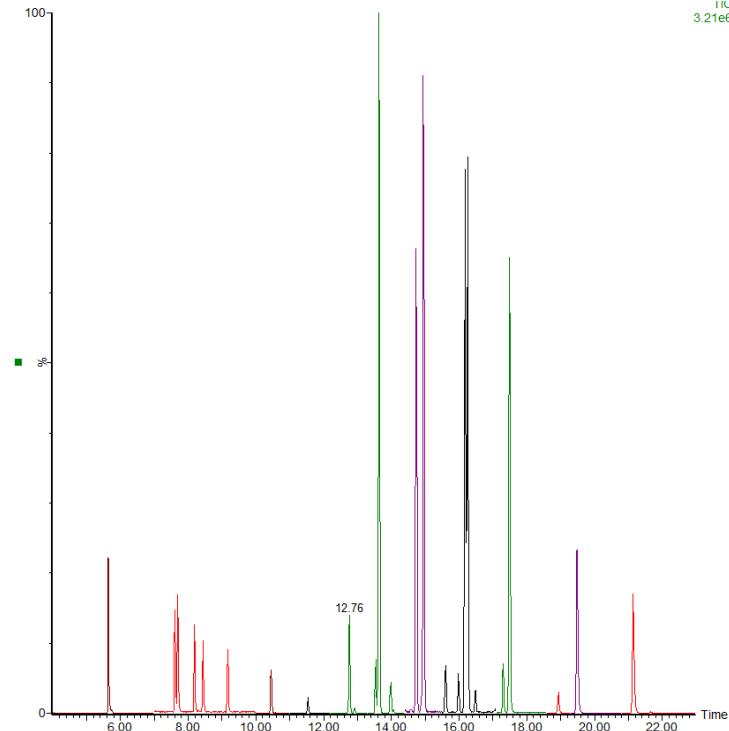
Chromatographie

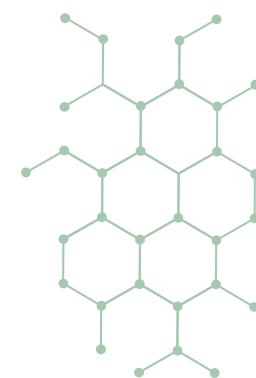
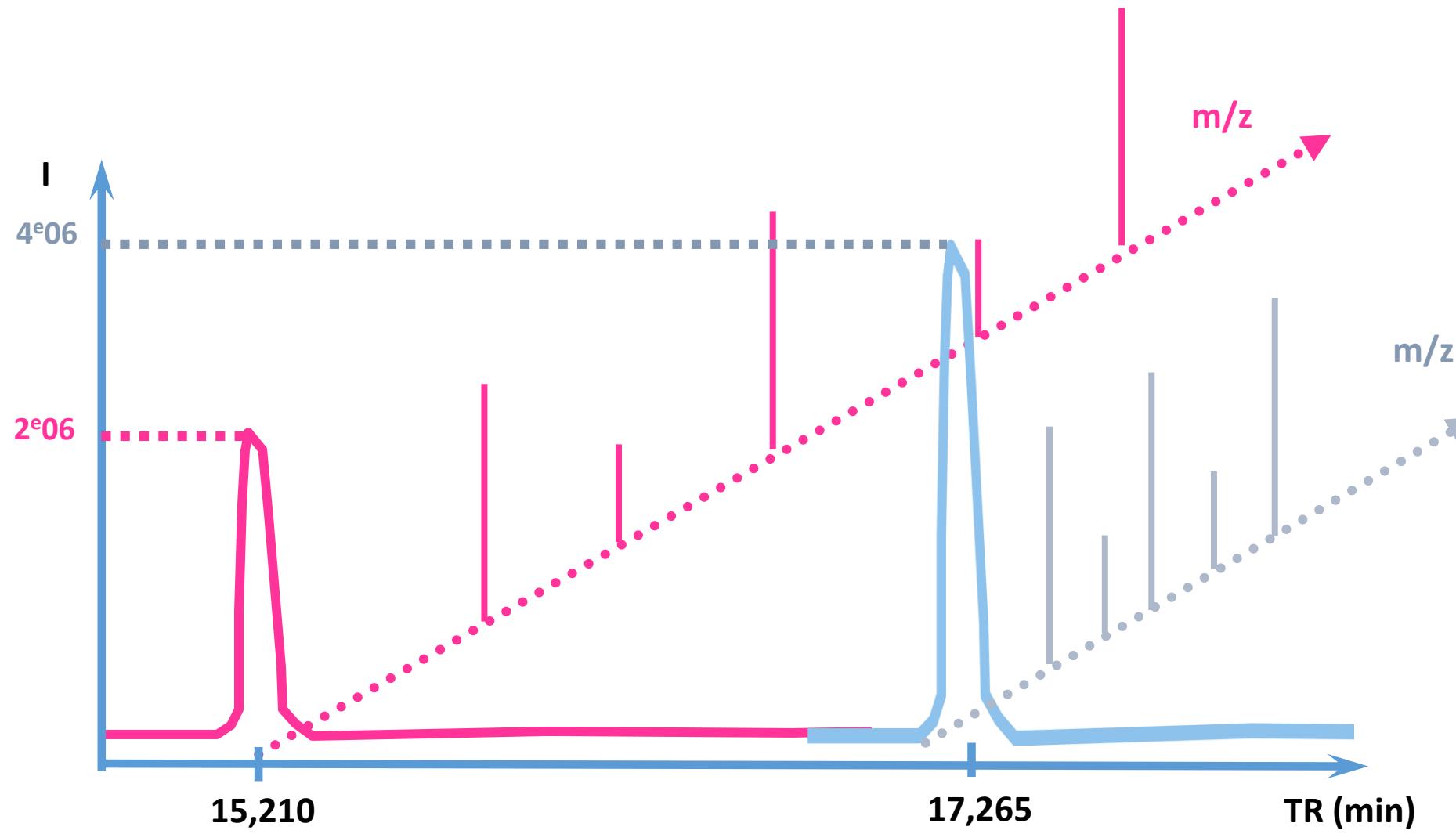


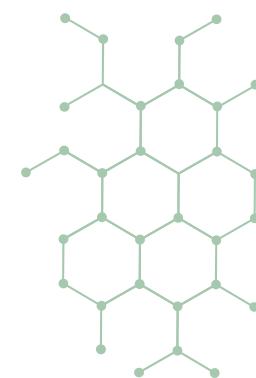
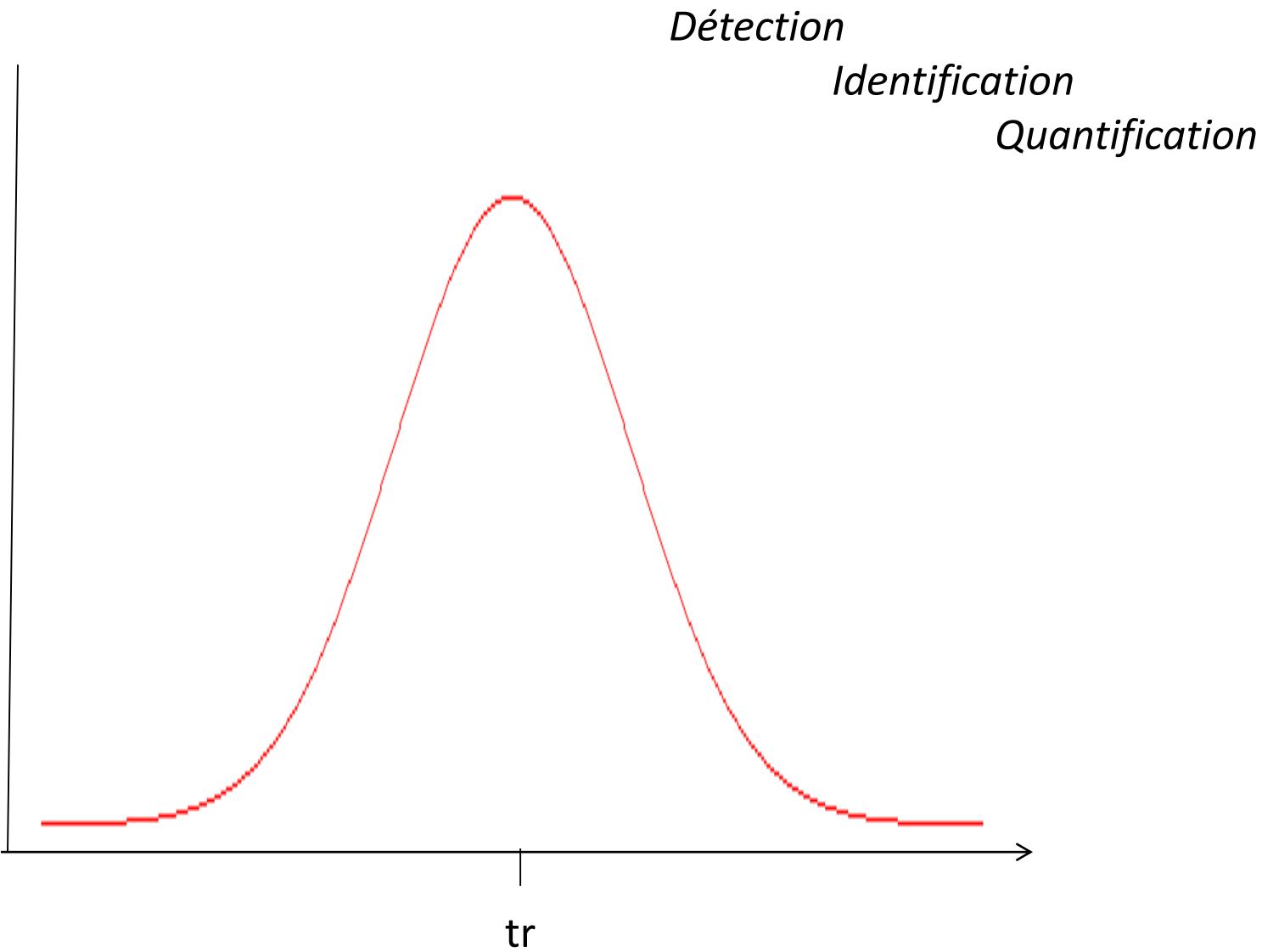
Matrices complexes
Sensibilité



<https://c-marketing.eu/7-etapes-pour-simplifier-un-contenu-complexe/>







UHPLC



Ionisation

Chromatographie



Source d'ionisation

ESI



Analyseur

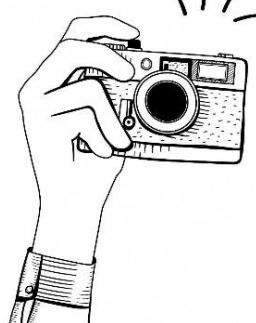
Quadripole



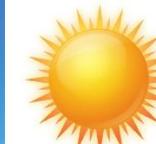
Détecteur



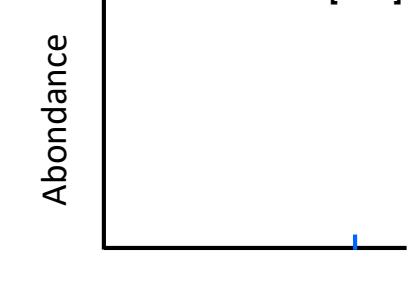
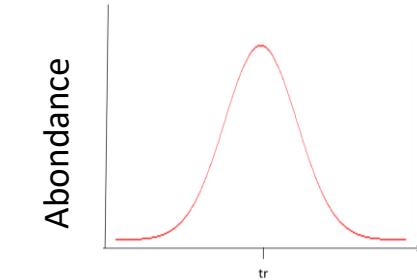
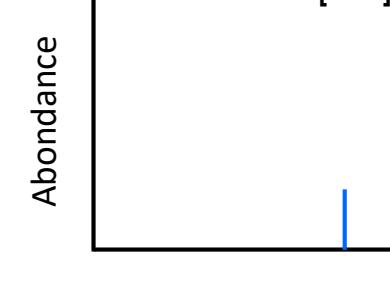
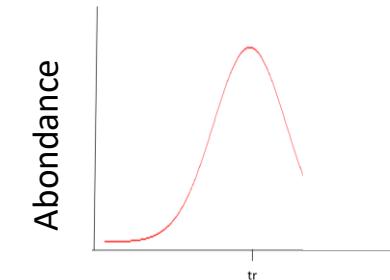
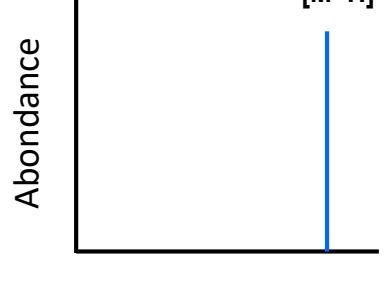
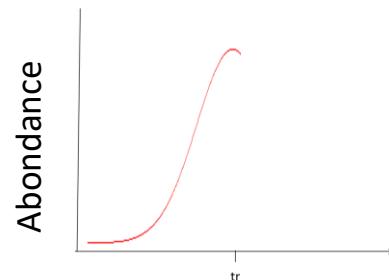
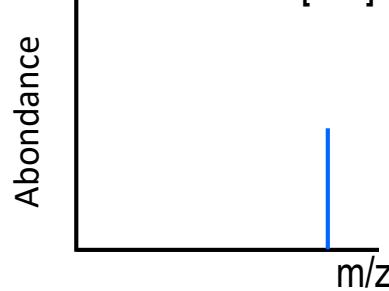
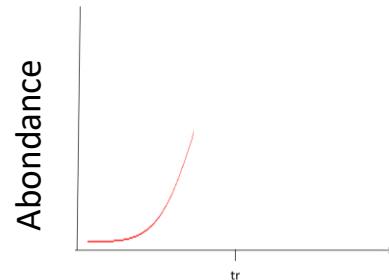
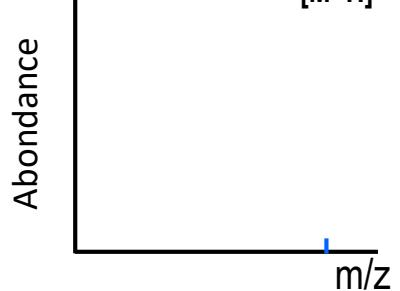
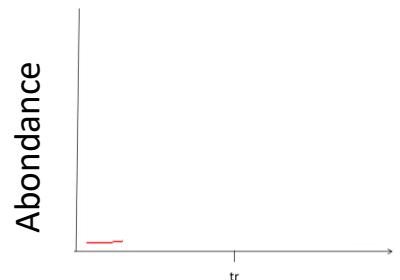
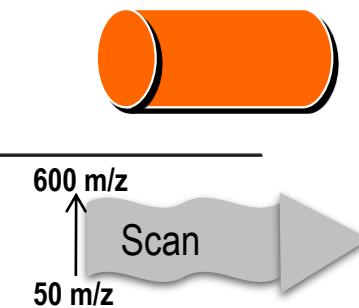
Spectrométrie de Masse



Chromatographie

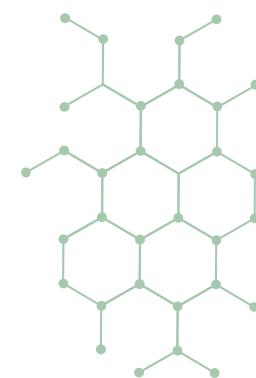
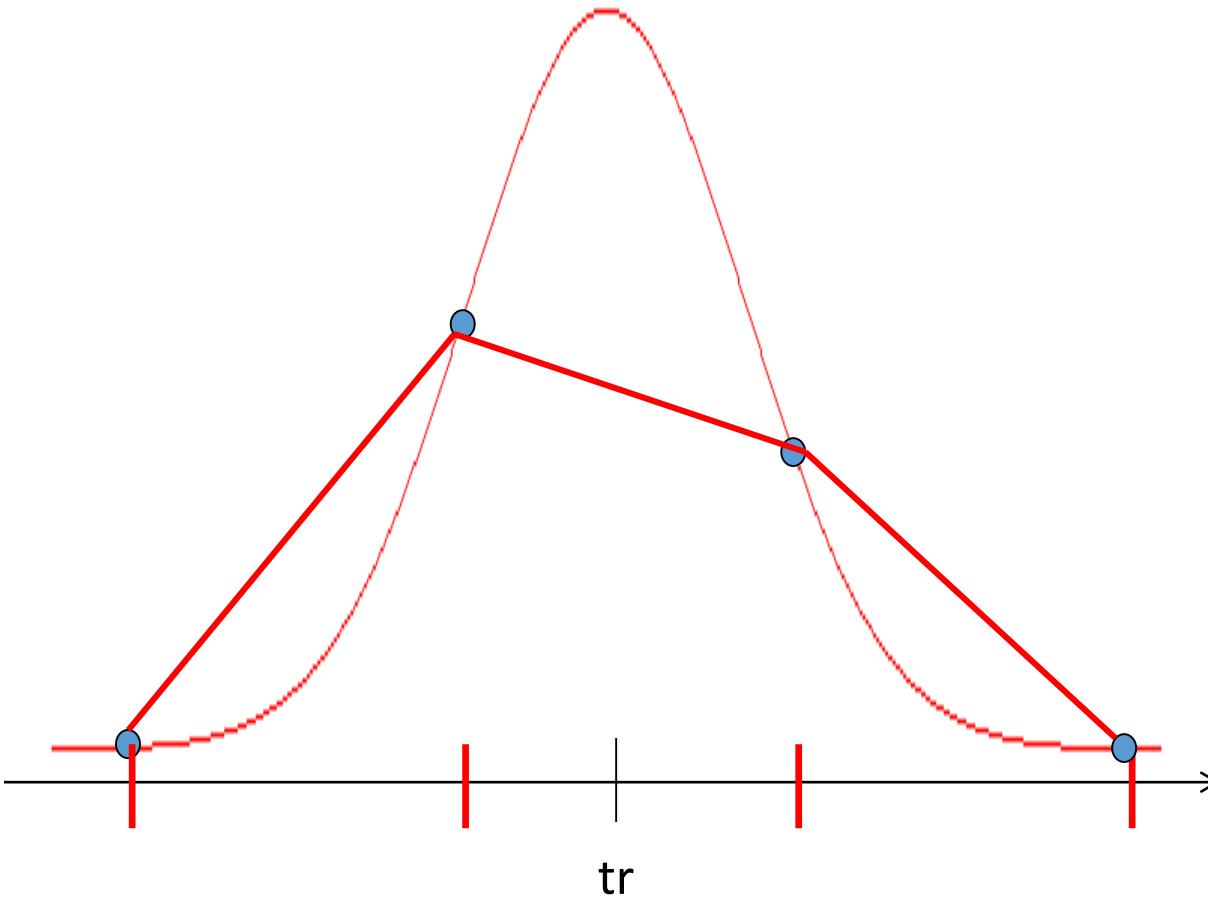


Ionisation

UHPLC**Chromatographie****Source d'ionisation****ESI** $[M+H]^+$ **Analyseur****Quadripole**

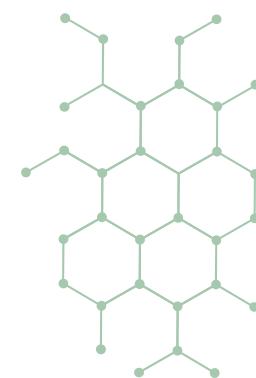
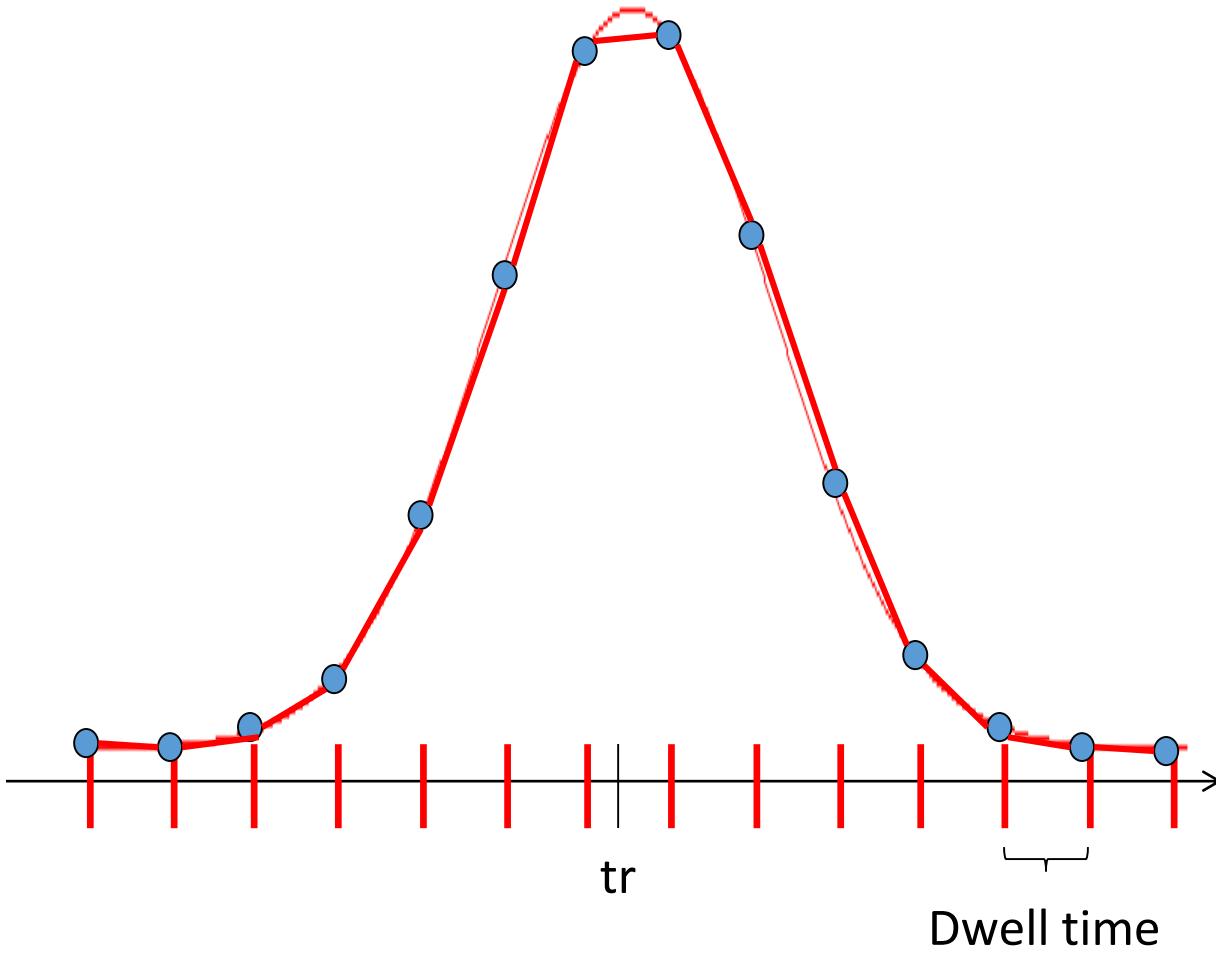
$[M+H]^+$ 

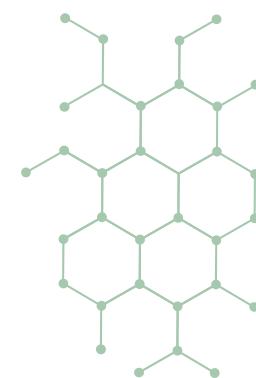
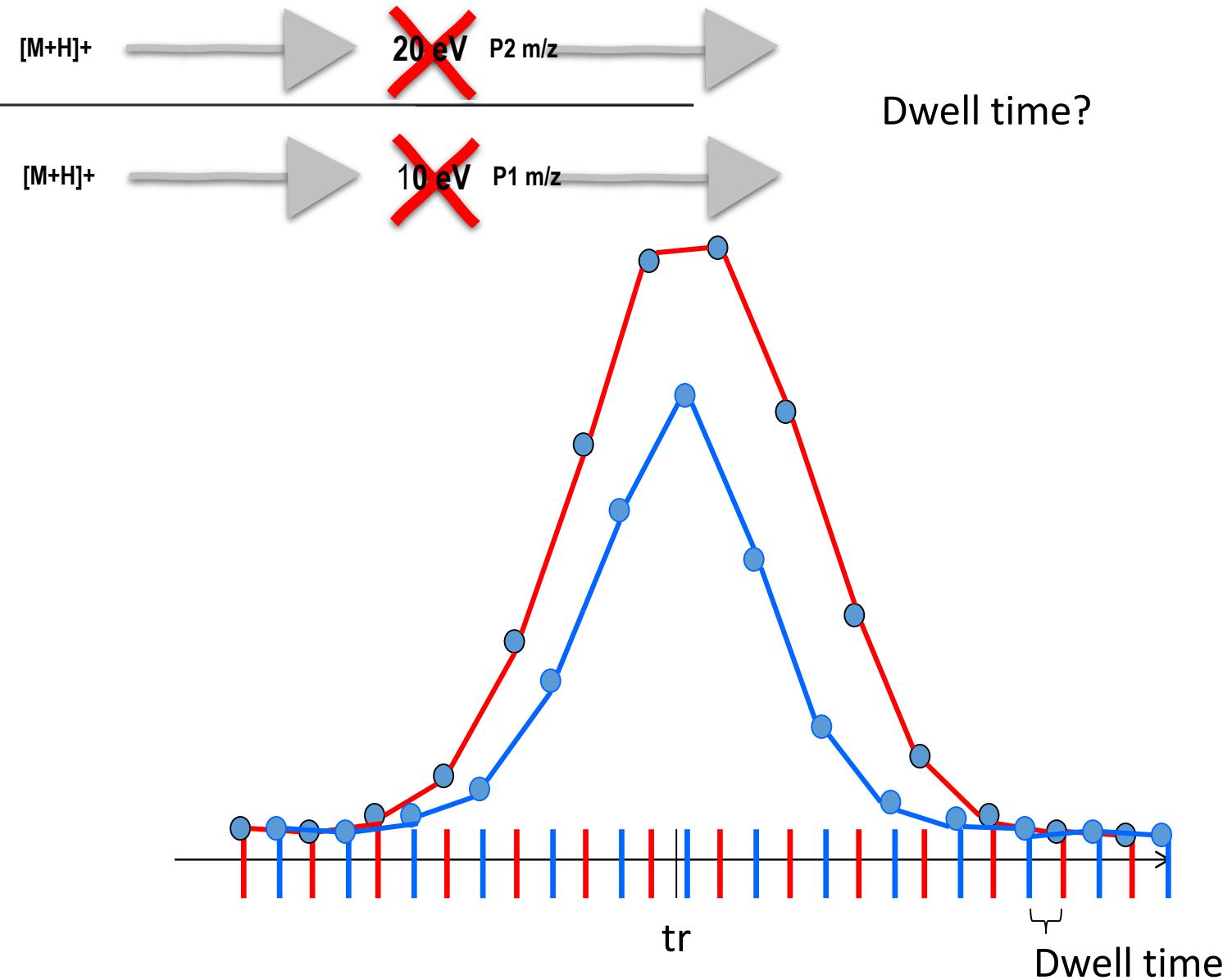
Pendant combien de temps ?

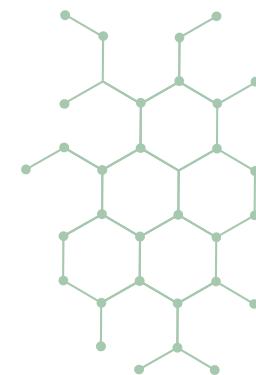
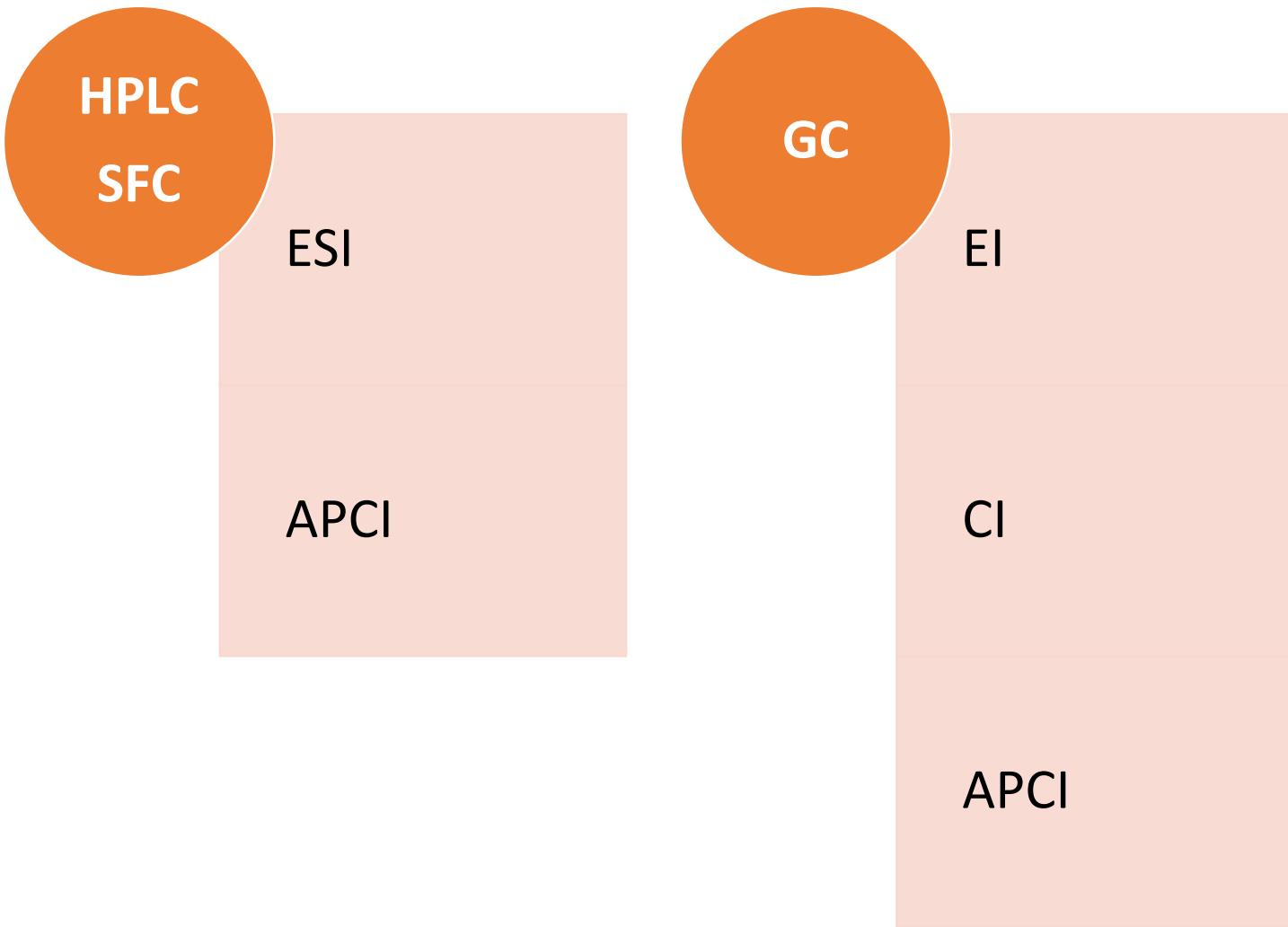


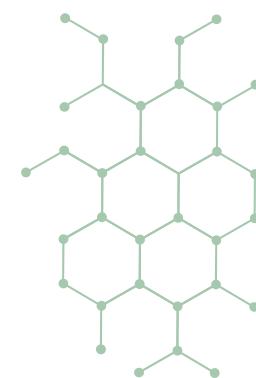
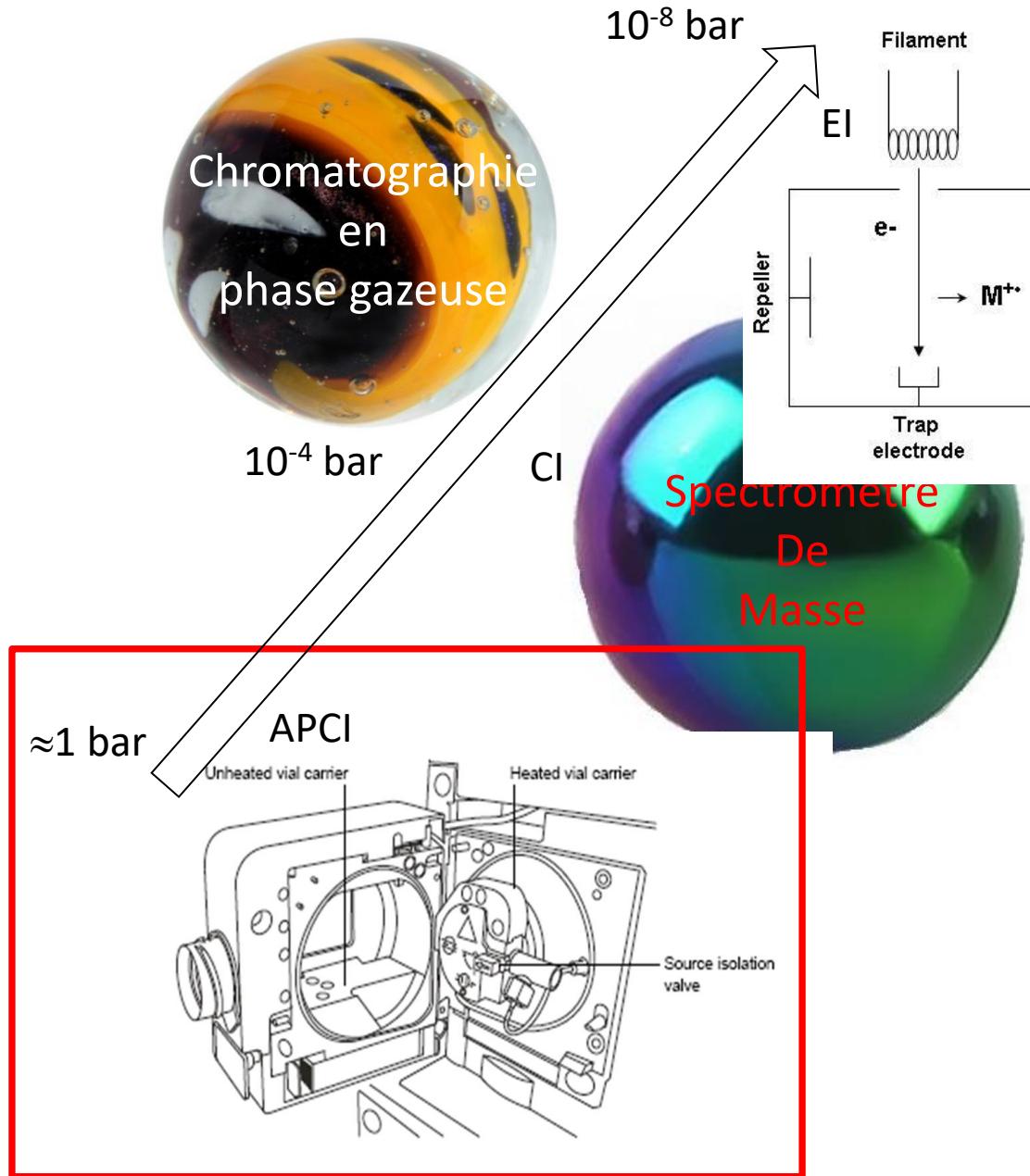
$[M+H]^+$ 

Pendant combien de temps ?





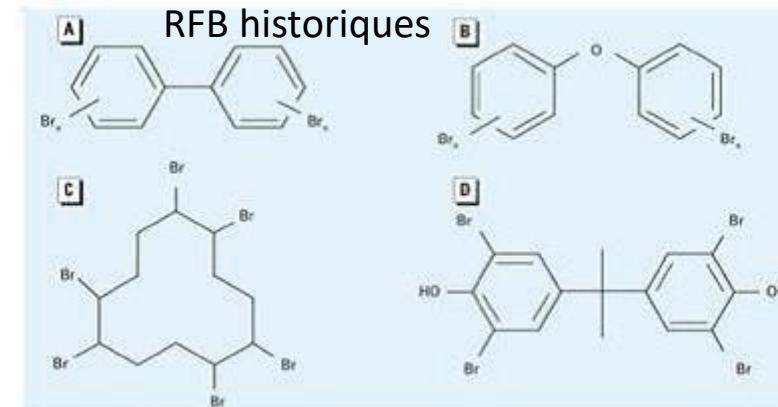




Substances chimiques **ignifuges** visant à inhiber ou retarder le processus de combustion.

Les retardateurs de flamme bromés (**RFB**) représentent **30%** des retardateurs de flammes utilisés en Europe

- Boîtiers électroniques et électriques (plastiques)
- Plaques de circuits imprimés, Fils et câbles
- Mobilier capitonné, matelas
- Tissus d'ameublement
- Transports (train, bateau, avion)
- Constructions
- Textiles
- Jouets

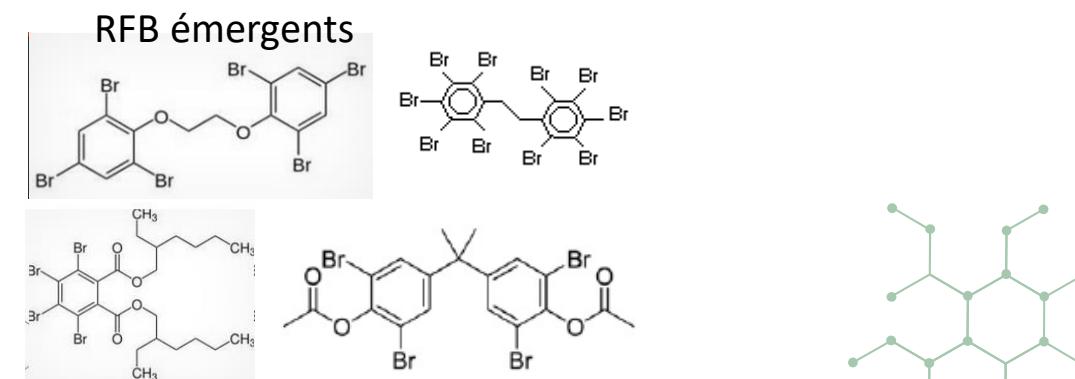


Contaminants de l'environnement , **bioaccumulables** dans la chaîne alimentaire

Au niveau alimentaire (données EFSA, 30 mai 2011)

- les poissons et autres fruits de mer ;
- les viandes et produits carnés ;
- les huiles végétales ;
- le lait et produits laitiers ;
- les oeufs et les ovo-produits.

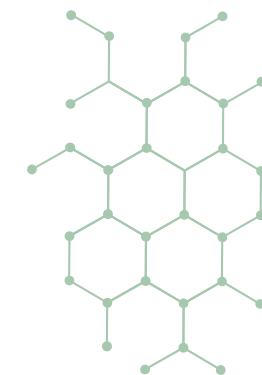
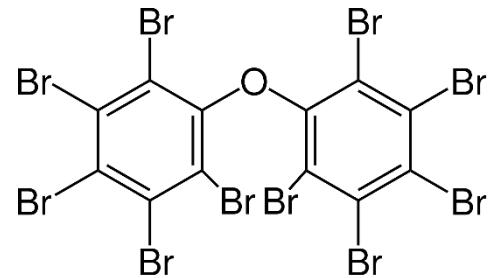
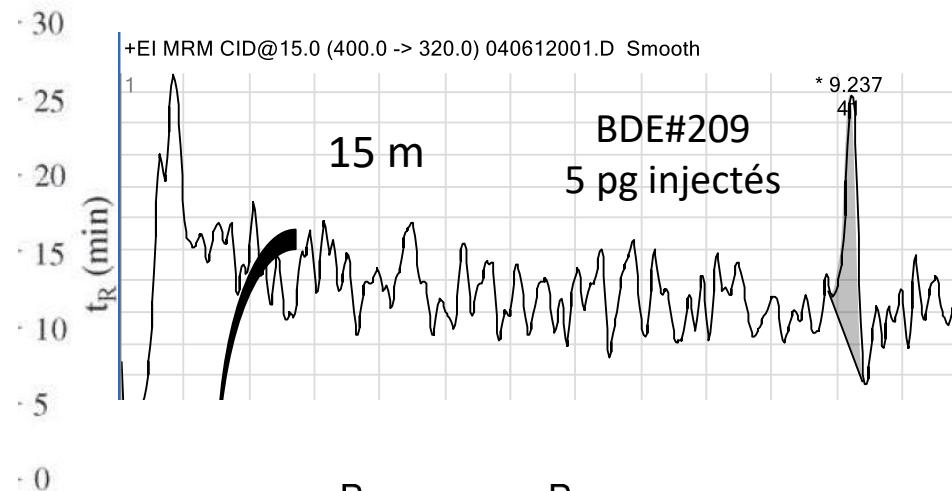
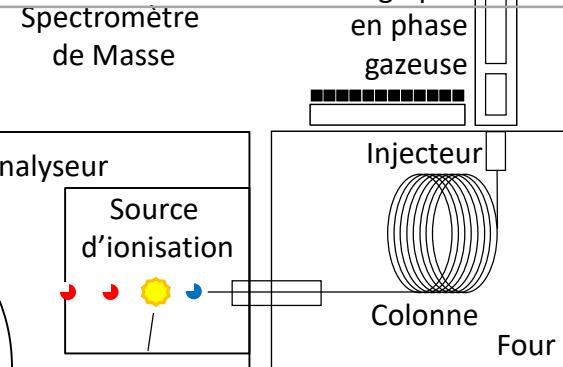
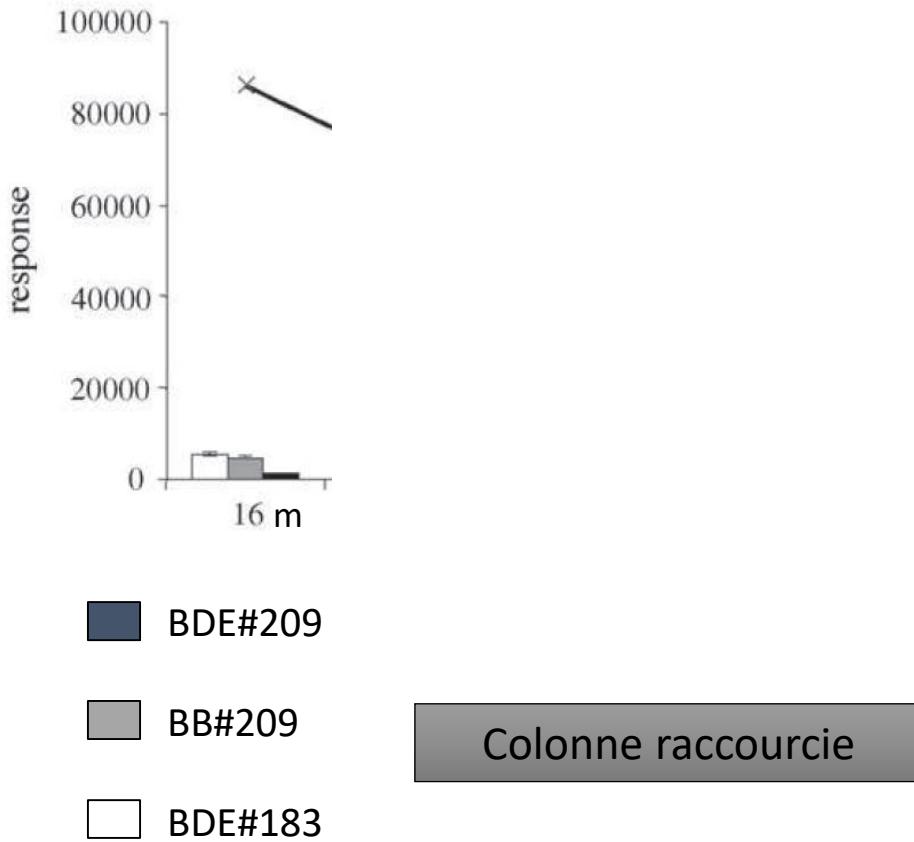
Différents types d'expositions : **alimentaire** et **occupationnelle**



Vers un gain en stabilité des RFB thermolabiles

Colonne Rtx-5MS (0.25 mm i.d.; 0.25 µm f.t.)

A. Binelli et al. / J. Chromatogr. A 1136 (2006) 243–247



GC/EI/HRMS (BE)



PBDE + nRFB

Run time : 45 min

DB5
MS

30 m x 0.25 mm, 0.25 µm

OBIND+DecaBDE

Run time : 7.25 min

Rtx
1614

GC/APCI/MS/MS

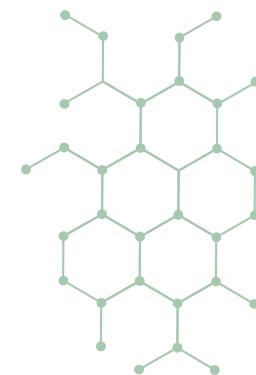


PBDE + nRFB + OBIND + DecaBDE

Run time : 13.6 min

Optima
5

2.5 m x 0.1 mm, 0.1 µm

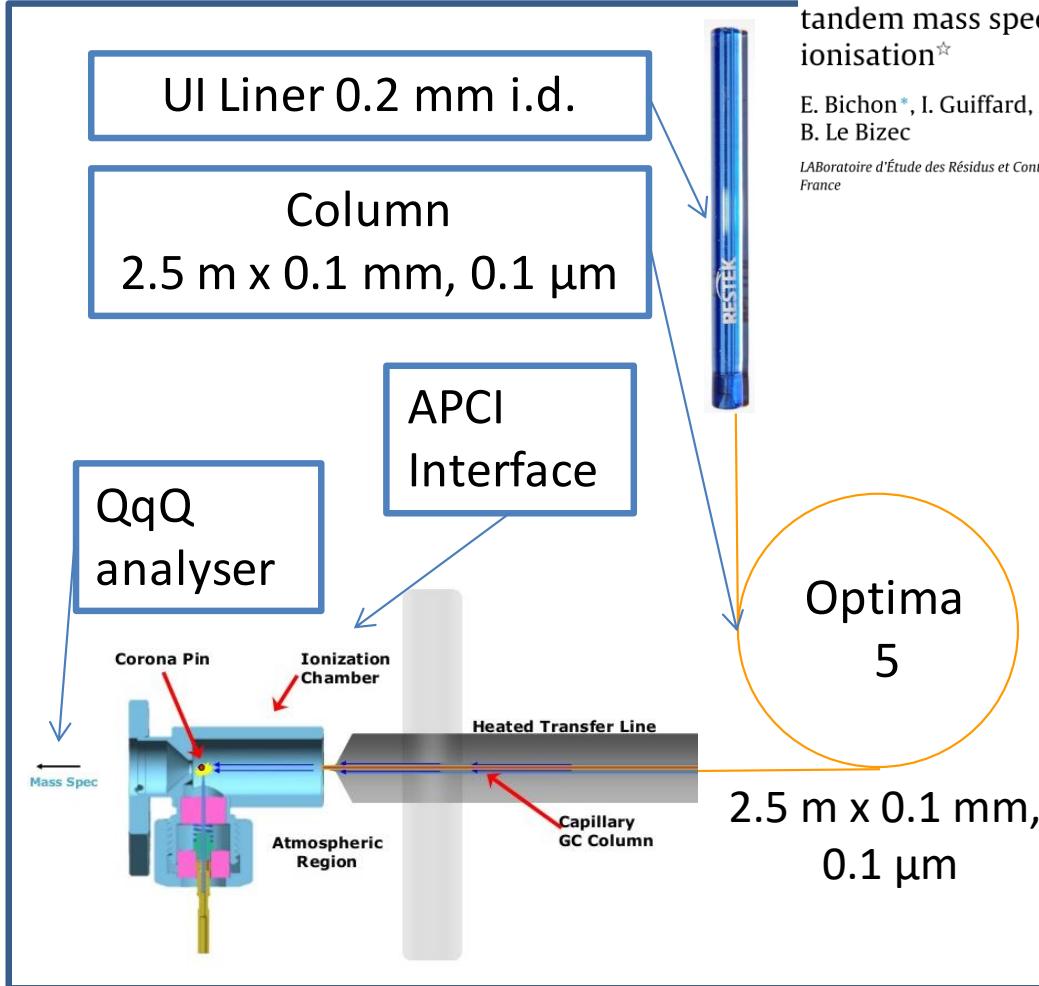




Simultaneous determination of 16 brominated flame retardants in food and feed of animal origin by fast gas chromatography coupled to tandem mass spectrometry using atmospheric pressure chemical ionisation*

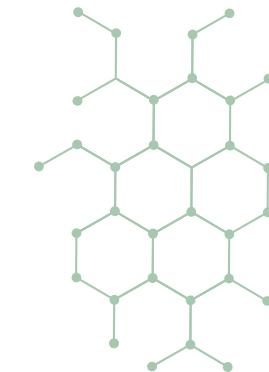
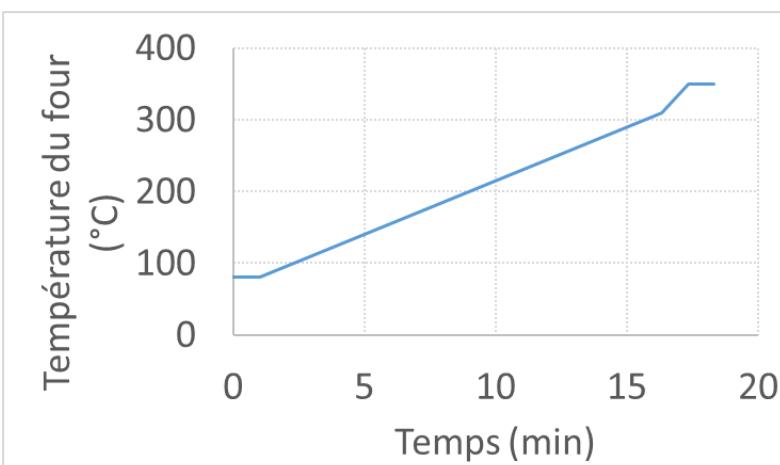
E. Bichon*, I. Guiffard, A. Vénisseau, E. Lesquin, V. Vaccher, A. Brosseaud, P. Marchand, B. Le Bizec

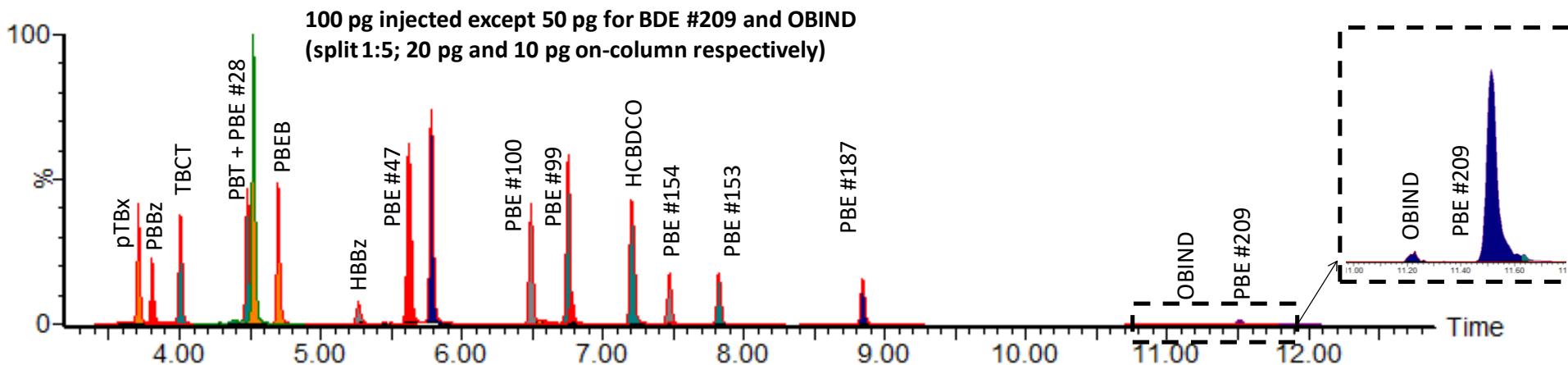
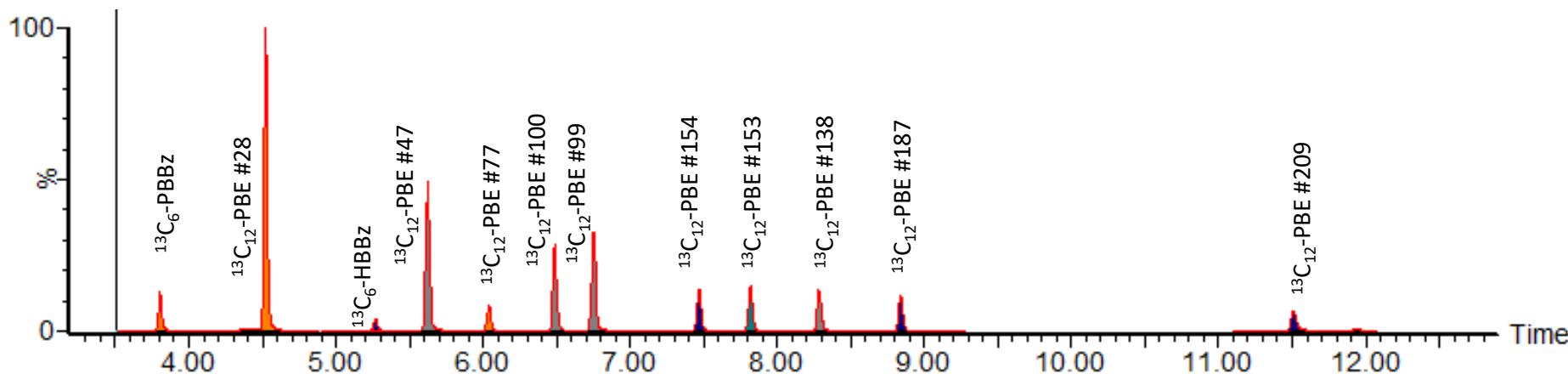
LABoratoire d'Etude des Résidus et Contaminants dans les Aliments (LABERCA), USC INRA 1329, Oniris, LUNAM Université, BP 50707, 44307 Nantes Cedex 3, France



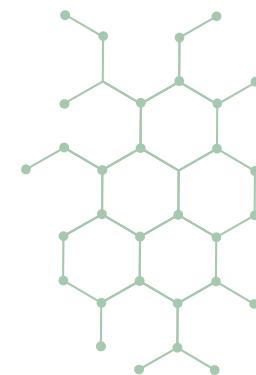
GC/APCI/MS

Plused Split mode (0,85 min @ 40 psi, split 1:5, 1 μ L injecté). T° injecteur: 275 °C Gaz vecteur (hélium) : debit constant (0.48 mL min⁻¹).

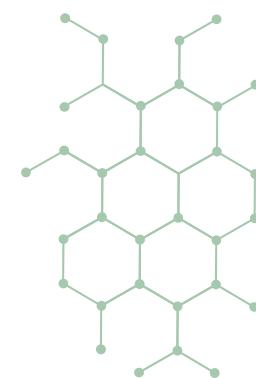
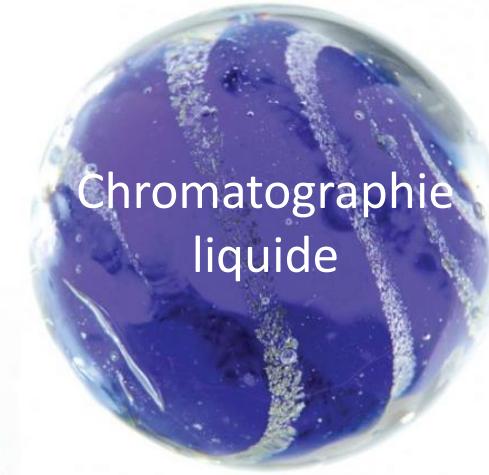


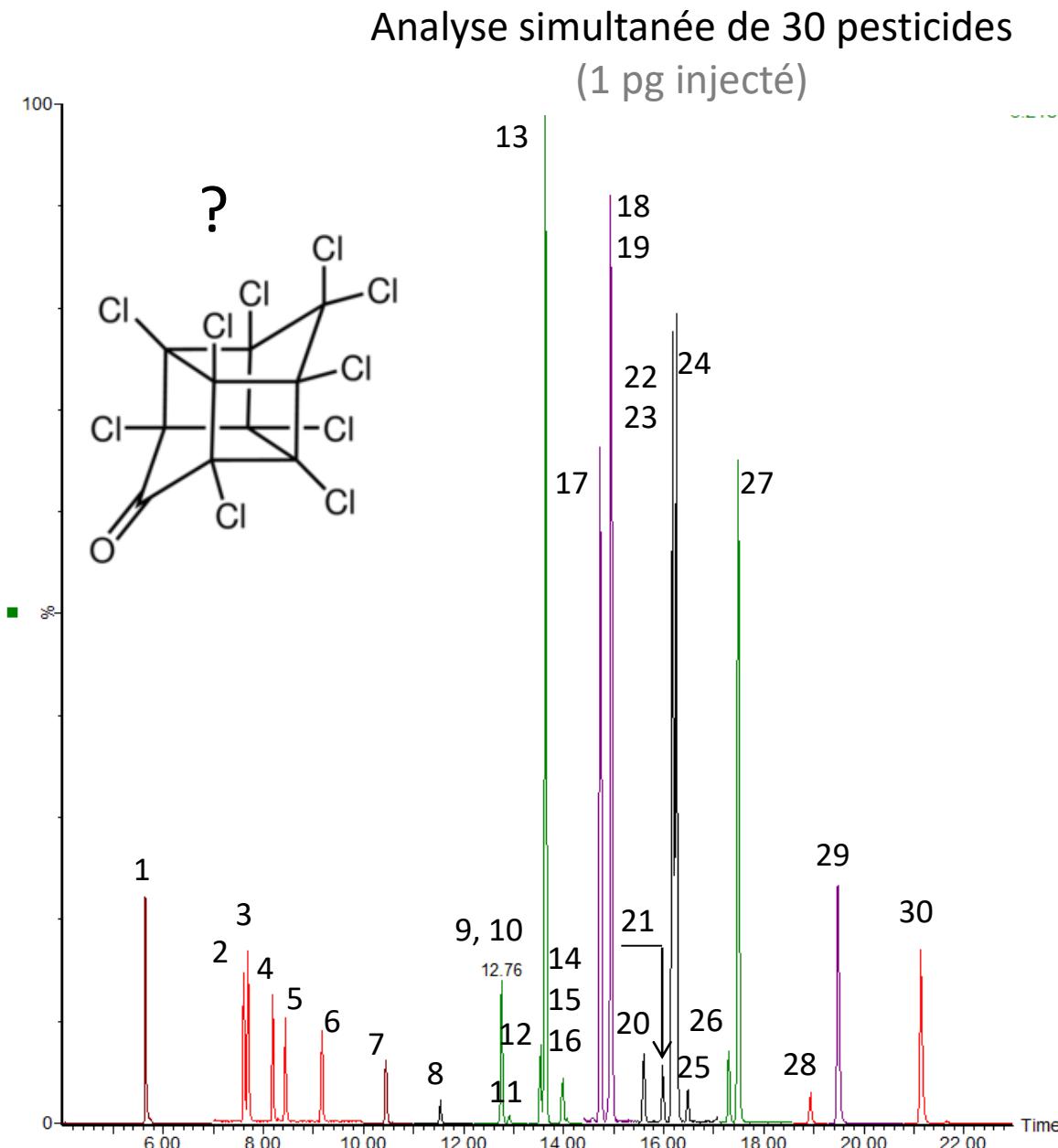


Chromatogramme d'ions extraits d'un mélange de 16 RFB natifs (en bas) et de leurs homologues marqués (en haut).

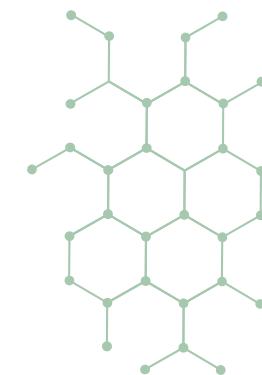


Limite de la GC/MS



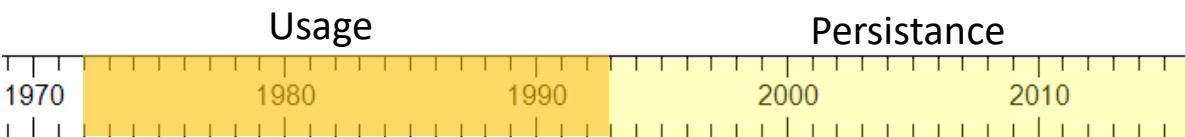


- 1) pentachlorobenzene
- 2) α -HCH
- 3) HCB
- 4) β -HCH
- 5) γ -HCH (=lindane)
- 6) δ -HCH
- 7) heptachlor
- 8) aldrin
- 9) oxychlordane
- 10) heptachlor epoxide cis
- 11) heptachlor epoxide trans
- 12) γ -chlordanne (=trans)
- 13) o,p'-DDE
- 14) α -chlordanne (=cis)
- 15) α -endosulfan
- 16) trans-nonachlor
- 17) p,p'-DDE
- 18) dieldrin
- 19) o,p'-DDD
- 20) endrin
- 21) β -endosulfan
- 22) cis-nonachlor
- 23) p,p'-DDD
- 24) o,p'-DDT
- 25) endrin aldehyde
- 26) endosulfan sulfate
- 27) p,p'-DDT
- 28) endrin ketone
- 29) methoxychlor
- 30) mirex (=perchlordecone)



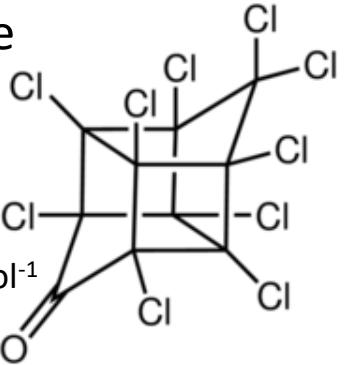
Problématique

Cas de la chlordécone

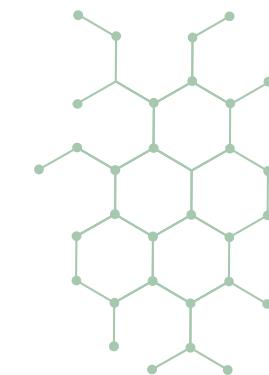
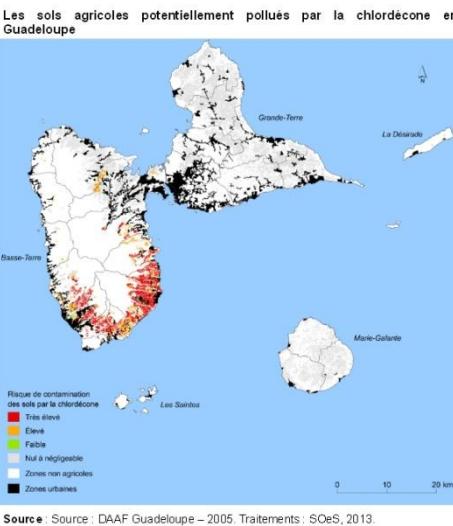
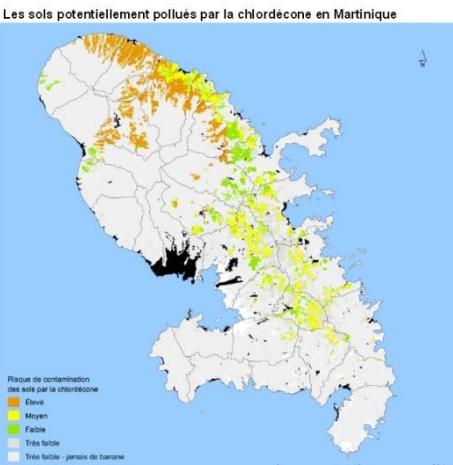


Chlordécone

$C_{10}H_{10}O$
MM = 490,6 g.mol⁻¹
Log K_{ow}=4,5



Sensibilité
Molécule difficile



RAPID COMMUNICATIONS IN MASS SPECTROMETRY

Rapid Commun. Mass Spectrom. 2004; **18**: 1243–1244Published online in Wiley InterScience (www.interscience.wiley.com)

RCM

Letter to the Editor

To the Editor-in-Chief
Sir,

Detection of chlordécone by liquid chromatography with tandem mass spectrometry

Chlordecone is a chlorinated hydrocarbon used in various industrial media and developed to complete GC. We have developed a method for chromatography-mass spectrometry (MS/MS) to simplify detection of chlordecone samples.

Chlordécone
Wako Pure Chemicals Inc.
(Osaka, Japan)

Journal of Chromatography A, 1408 (2015) 169–177

Contents lists available at ScienceDirect

Journal of Chromatography A

journal homepage: www.elsevier.com/locate/chroma

ELSEVIER

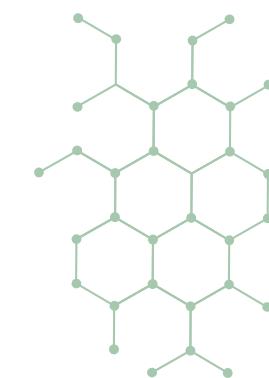
Ultra-trace quantification method for chlordécone in human fluids and tissues

Emmanuelle Bichon ^{a,*}, Ingrid Guiffard ^a, Anaïs Vénisseau ^a, Philippe Marchand ^a, Jean-Philippe Antignac ^{a,b}, Bruno Le Bizec ^a

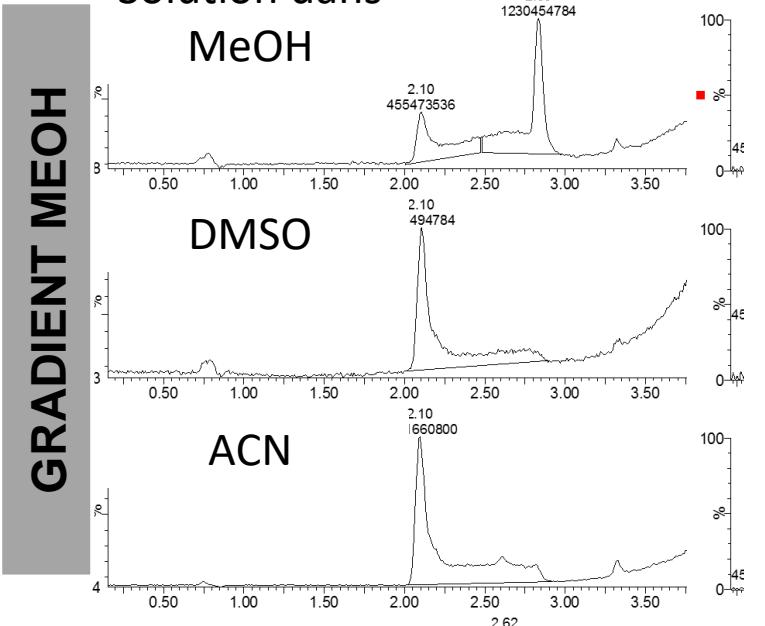
^a LUNAM Université, Oniris, Laboratoire d'Etude des Résidus et Contaminants dans les Aliments (LABERCA), Nantes F-44307, France

^b INRA, Nantes F-44307, France

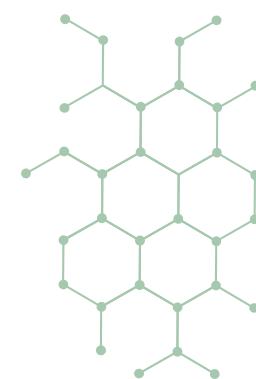
GC column (l. in m, i.d. in mm, f.t.in μm)	Instrument	Pitfalls
RTX-1614 (15/0.25/0.10)	GC-EI-HRMS (JEOL 800D)	Coelution not studied
DB5MS (30/0.25/0.25)	GC-APCI-MS/MS (WATERS APGC)	peak tailing on chlordécone coelution on 3 couples of OCPs
HT8-PCB (60/0.25/0.25)	GC-EI-HRMS (JEOL 800D)	peak tailing on chlordécone coelution chlordécone and cis-nonachlor
Optima-17MS (30/0.25/0.25) and (30/0.25/0.15)	GC-EI-MS (Agilent 5973)	peak tailing on chlordécone coelution chlordécone and endrin



Solution dans
MeOH



Colonne C30 Accucore



Préparation d'échantillons

Adipose tissue
(1 g)

Breast milk
(3 mL)

Serum
(1 mL)

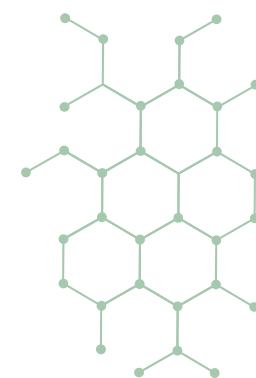
PLE
tol/acetone

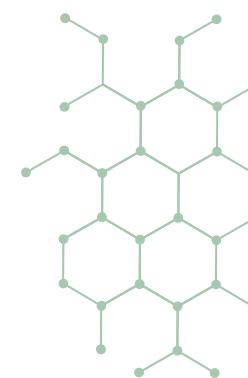
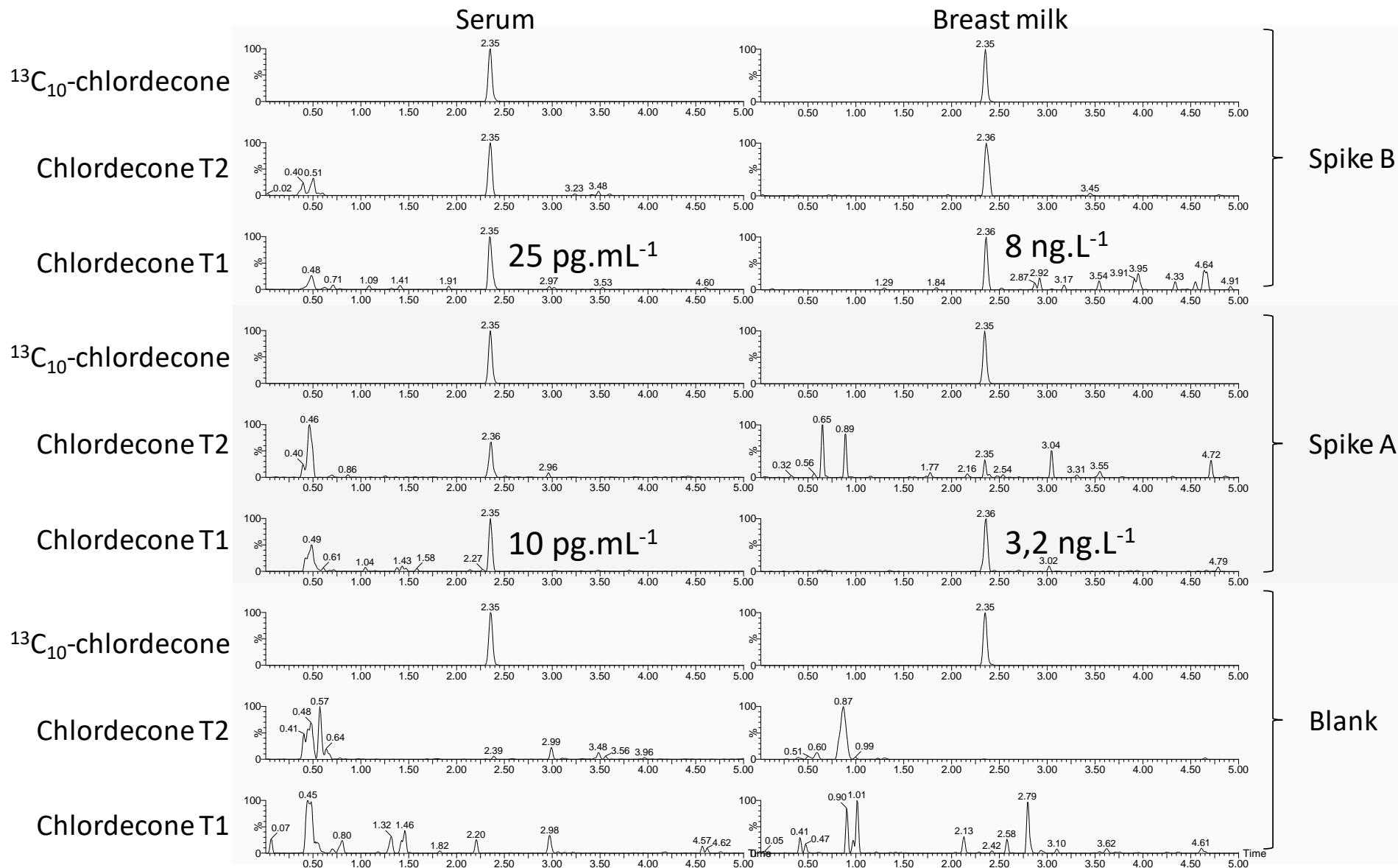
LLE
EtOH/ether/Hex

SPE
ACN elution

GPC

LC-(ESI)-MS/MS
Accucore C30 (Thermo) 100 x 2 mm, 2.6 µm
Transitions 506.7>426.7 and 508.7>428.7 @ 20 eV





Quelle chromatographie associée à la spectrométrie de masse ?



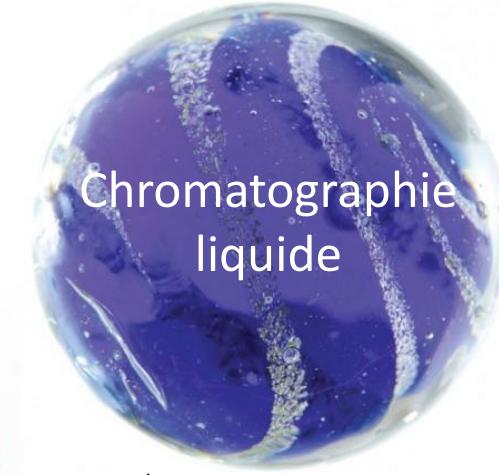
Chromatographie
en
phase gazeuse



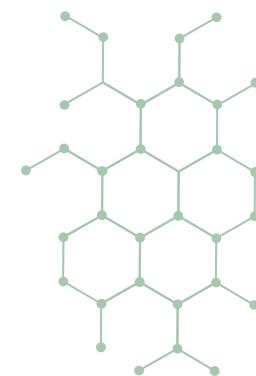
Spectromètre
De
Masse



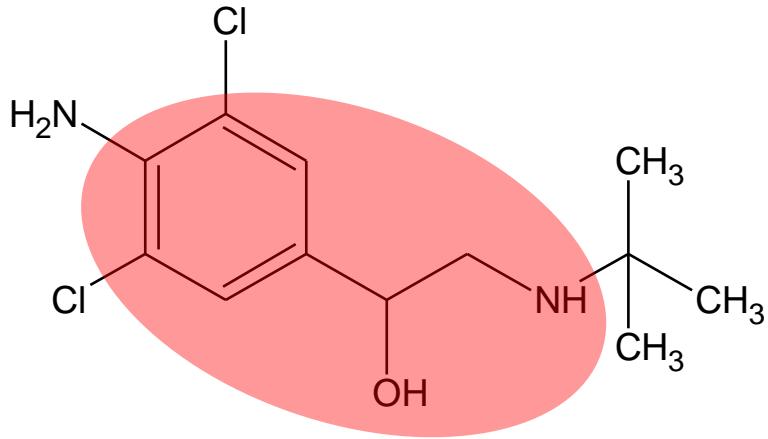
Chromatographie
par fluide
supercritique



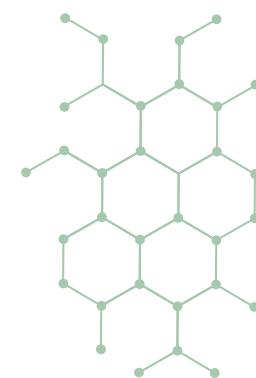
Chromatographie
liquide

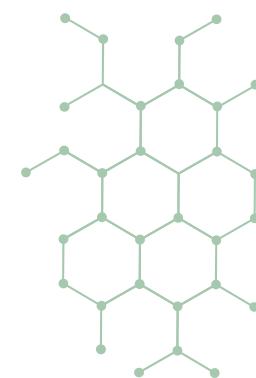
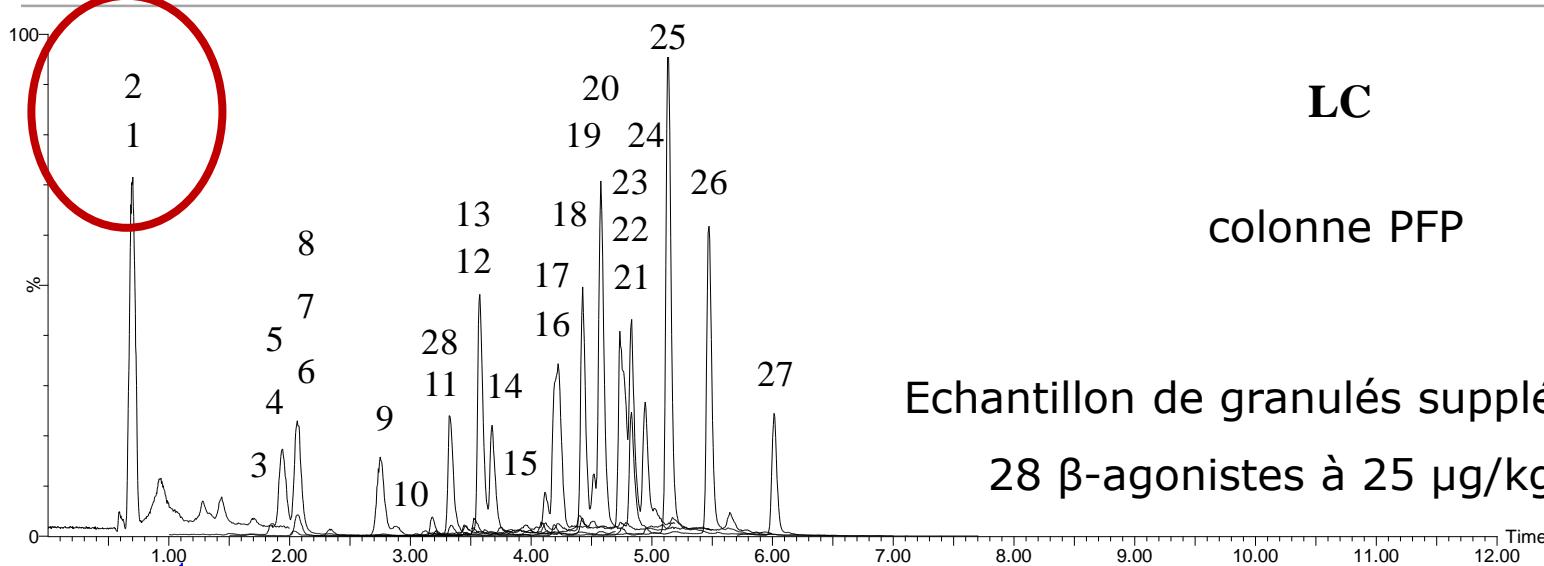


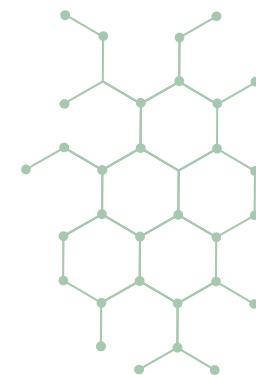
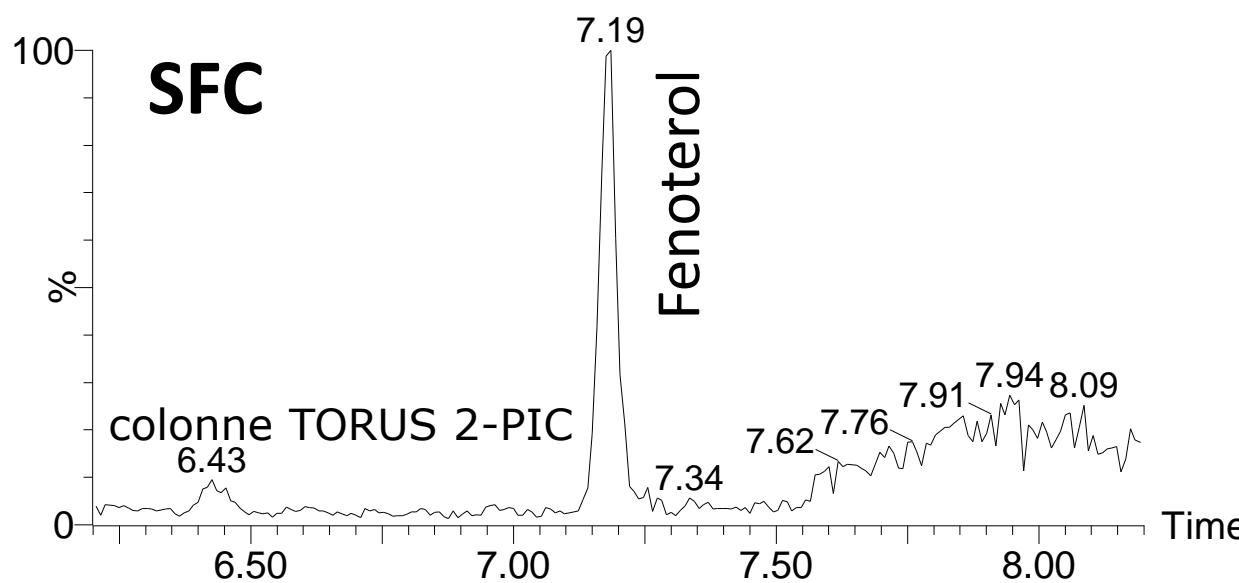
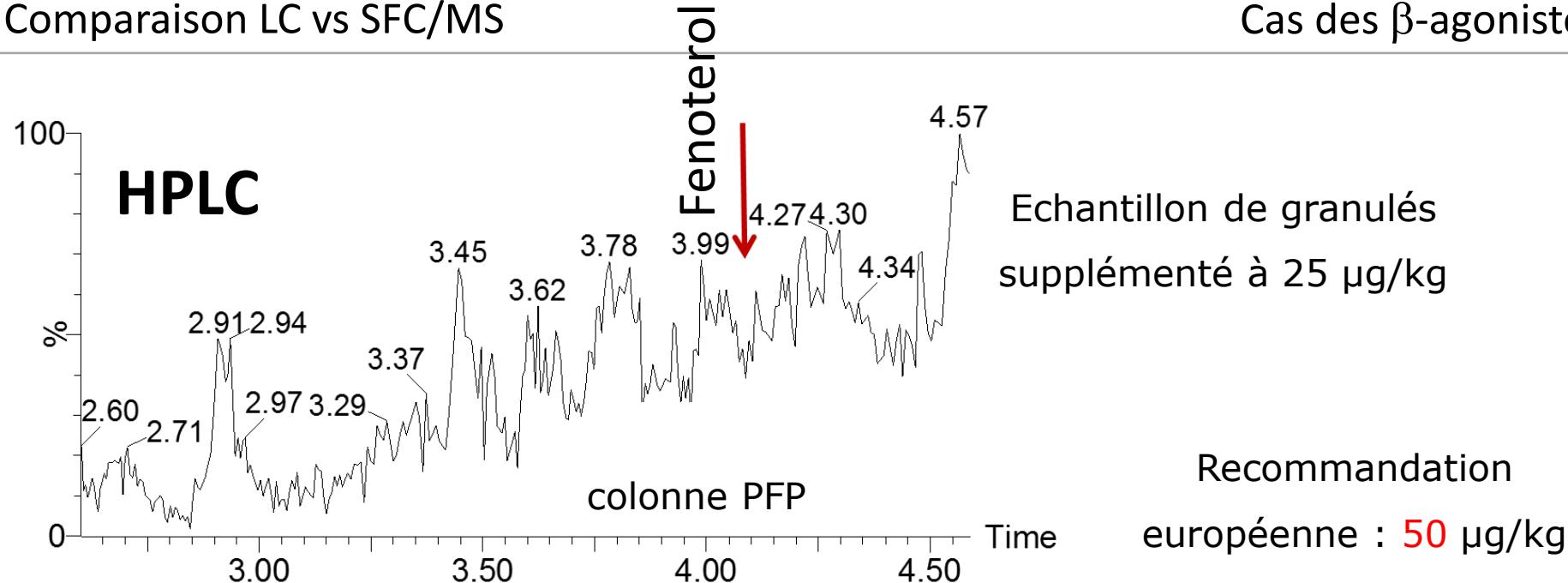
Promoteurs de croissance en élevage

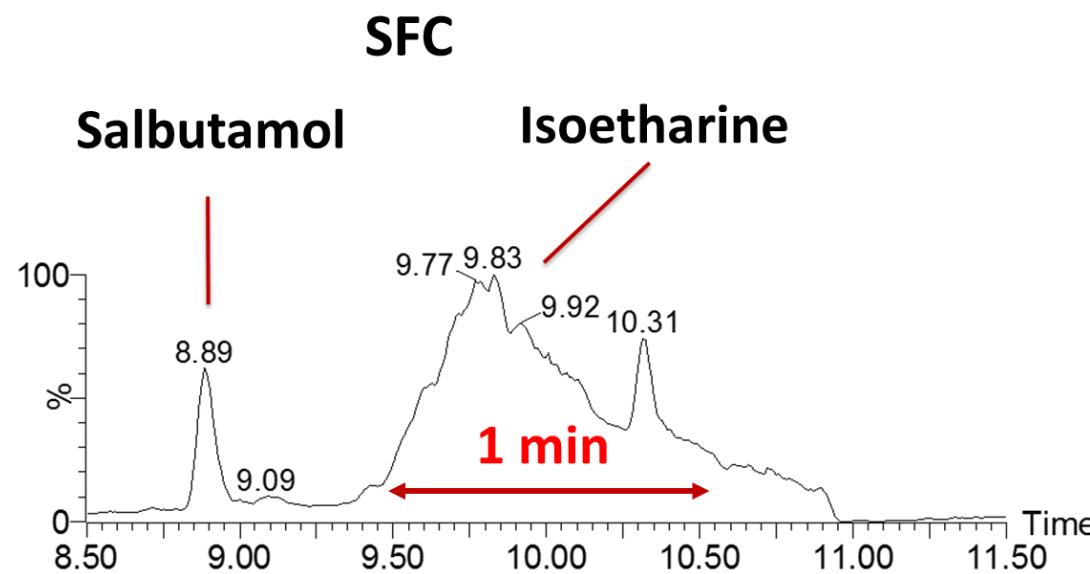
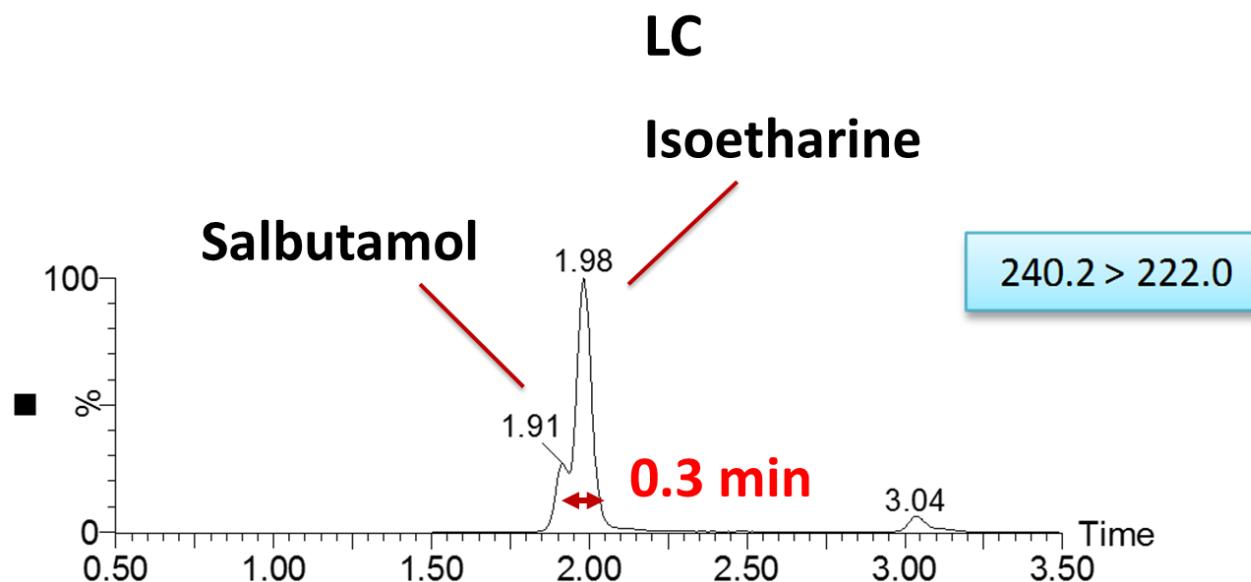


Substances interdites
(Annexe A directive 96/22/EC)



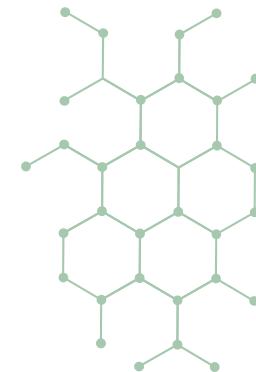






Gain en Séparation

Au détriment de la résolution pour l'isoétharine...



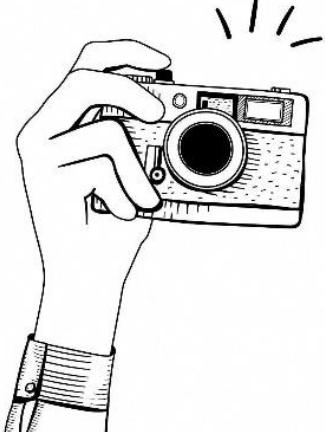
Compromis entre résolution/sensibilité chromatographique et résolution/sensibilité MS

Attention au choix de l'ionisation = visualiser les analytes



Ionisation

Spectrométrie de Masse

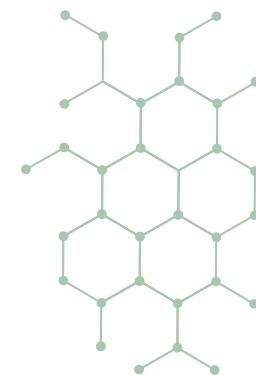


Chromatographie



COUPLAGE PERTINENT POUR :

- l'analyse d'ultratraces
- la sélectivité
- les matrices complexes
- la quantification
- le Non-Target Screening





Merci pour votre attention